Tips & Tricks for improved quantitative and qualitative LC-MS



Overview

This work describes various measures and options for maximized LC-MS sensitivity and low Limit of Detection (LOD) by avoiding contaminations causing signal suppression, adduct formation and increased background noise or spectrum complexity:

- Selection and handling of HPLC columns
- Choice and cleaning of laboratory equipment (bottles, vessels, funnels etc.)
- Choice and proper processing of solvents and additives

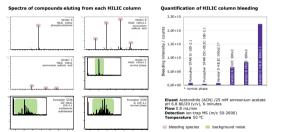
HPLC columns - Low bleed

In LC-MS analyses, column bleed (loss of stationary phase entities) leads to signal suppression, adduct formation and sensitivity drop. Two options are available to minimize these negative effects:

- Utilization of low-bleed stationary phase material
- Implementation of a column flushing/washing process

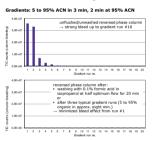
Low bleeding ZIC®-HILIC columns

Column bleeding studies utilizing LC-ion trap MS setup



Column flushing/washing

Removal of unbound organic entities

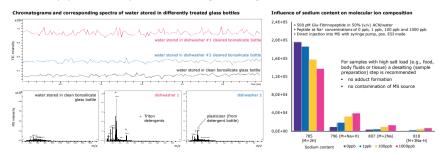


Laboratory equipment - Choice and cleaning

No dishwashing of glass vessels; strong bases and surfactants lead to contamination of eluents by
• Dissolution of silica and alkali under aggressive cleaning conditions, formation of adducts likely

• Deposition of detergents (from cleaning process) and plasticizers (from detergent bottle) on glass surface

Clean all equipment via evaporation or flushing with MS-grade solvents



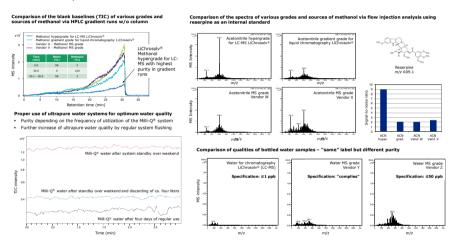
The life science business of Merck operates as MilliporeSigma in the U.S. and Canada.

SigmaAldrich.com/LC-MS

Solvents and additives - Purity, handling and storage

Purity

- Utilization of MS-grade solvents and ultrapure additives is mandatory
- Avoid any contamination during storage and handling
- Additives (formic or acetic acid, NH₄OH, triethylamine etc.) must be volatile, nonvolatile salts precipitate in MS
- Buffers ionize analyte molecule M, but ammonium, <u>formate</u> or <u>acetate</u> also form adducts [M+buffer]
- Choose eluent pH in accordance with your stationary phase properties (e.g., for silica pH 2-7.5 is recommended)



Handling & Storage

- Standard glass bottles: Silica and alkali dissolution, formation of adducts with analytes
- Plastic devices (bottles, funnels etc.): solvent contamination by leaching of additives (plasticizers, stabilizers, anti-static agents), accumulation on stationary phase and elution as ghost peaks likely Bottle caps and adapters:
- Professional equipment directly mounted to the original brown glass bottle no homemade solutions; avoid decanting

