# A Comparison of Porous Layer Open Tubular (PLOT) Columns for the Separation of Refinery Gases

TracePLOT Alumina Columns

as sodium sulfate or potassium chloride to control retention

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Alumina is often used for the analysis of volatile hydrocarbons due to its selectivity which

retentive nature of alumina requires that the surface be deactivated with inorganic salts such

provides baseline resolution of most isomers at above ambient temperatures. The bighty

Note the difference in selectivity between the alumina deactivated with Na<sub>2</sub>SO<sub>4</sub> (Figure 1)

versus the alumina KCI (Figure 2) deactivation. For example alumina Na<sub>2</sub>SO<sub>4</sub> elutes

#### Overview

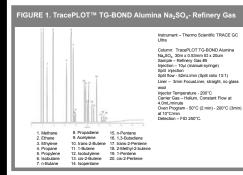
This poster demonstrates the use of various Thermo Scientific TracePLOT Porous Layer Open Tubular columns for the analysis of low molecular weight hydrocarbons.

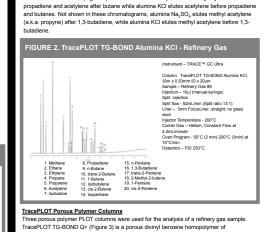
#### Introduction

Porous Layer Open Tubular (PLOT) columns are well suited for the analysis of light hydrocarbons such as those found in refinery gases. These highly selective columns are capable of separating low molecular weight hydrocarbons at above ambient temperatures and the columns can then be programmed to higher temperatures to elute higher boiling compounds. An increasingly wide variety of PLOT column coatings are available, including alumina deactivated with Na<sub>2</sub>SO<sub>4</sub> and with KCI, as well as a range of porous polymers with varying degrees of polarity.

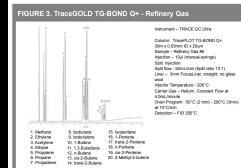
The differences in selectivity of numerous types of Thermo Scientific TracePLOT columns are demonstrated by the differences in the separation of light hydrocarbons in a refinery gas sample.

#### Results

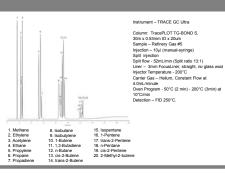


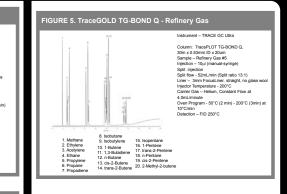


TracePLOT TG-BOND Q+ (Figure 3) is a porous divinyl benzene homopolymer of intermediate polarity incorporating a lower around +-vinyl pyridine into the polymer. TracePLOT TG-BOND S (Figure 4) is a midpolarity divinybenzene incorporating higher amount of 4-vinyl pyridine. TracePLOT TG-BOND Q (Figure 5) is a non-polar 100% divinybenzene polymer. These three different types of porous polymers show significant changes in elution order as compared to the two alumina PLOT columns. This would be useful for purtly analysis of a specific hydrocarbon in situations where an impurity elutes after the main peak of interest on an alumina column, which can cause in the impurity to be overrun by the main peak, resulting in co-elution. The differences in selectivity between these columns does not result in elution order changes between these three porous polymer PLOS columns, but there are differences in separation factor for many of the components.



### FIGURE 4. TraceGOLD TG-BOND S - Refinery Gas





#### Conclusions

 Separation of saturated, unsaturated and branched chain light hydrocarbons such as those found in refinery gases is best accomplished on deactivated alumina PLOT columns.

 Porous polymers PLOT columns offer a different selectivity and elution order for refinery gas components versus alumina which may be useful for purity analysis of individual hydrocarbons.

For additional information, please visit our Chromatography Resource Centre which can be found at: www.thermoscientific.com/chromatography

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