

Optimized Volatiles Analysis Ensures Fast VOC Separations

By Michelle Misselwitz, Gary Stidsen, and Chris English

- Optimized analysis allows for 36 runs per 12-hour shift, increasing instrument productivity.
- Rxi[®]-624Sil MS column inertness gives sharper peaks and more accurate data.
- High temperature stability reduces bleed profile, resulting in lower detection limits.

Optimized methods for the analysis of volatile organic compounds (VOCs) can be time-consuming to develop because compound lists can be extensive and analytes vary significantly in chemical characteristics. For example, target compounds in EPA Method 8260 for solid waste matrices include volatiles that range from light gases (Freon[®]) to larger aromatic compounds (trichlorobenzenes). These differences make column selectivity, thermal stability, and inertness critical to resolving volatiles. Often, "624" type columns are chosen for their selectivity, but thermal stability is usually poor, which can result in phase bleed that decreases detector sensitivity. New Rxi[®]-624Sil MS columns offer reliable resolution of critical VOC pairs and also provide lower bleed and greater inertness than other columns. In order to provide optimized conditions for labs analyzing VOCs, we established parameters that ensure good resolution, while reducing downtime by syncing purge and trap cycles with instrument cycles. In addition, we present comparative data that demonstrate why Rxi[®]-624Sil MS columns are the best choice for volatiles analysis.

Resolve Critical Pairs and Reduce Downtime

In order to achieve desired separations and minimize downtime between injections, several critical pairs were chosen for computational modeling using Pro *ez*GC software. The temperature program initially determined by the software was 35 °C (hold 5 min.) to 120 °C @ 11 °C/min. to 220°C @ 20 °C/min. (hold 2 min.). While this provided the best resolution of critical pairs, it also extended the analysis time to 19 min. Since the purge and trap cycle time was 16.5 min., we tested other conditions to see if adequate resolution could be maintained, while using a faster instrument cycle time that more closely matched the purge and trap cycle time, in order to maximize sample throughput. In other calculations, the software suggested changing temperature ramps at 60°C; therefore, a program of 35°C (hold 5 min.) to 60°C @ 11 °C/min. to 220°C @ 20 °C/min. (hold 2 min.) was tested. This final program reduced instrument downtime by better synchronizing injection and analysis cycles, and also provided excellent resolution of volatile compounds (Figure 1). Testing of faster conditions determined that the initial hold of 5 minutes at 35°C was critical for the best separation of early eluting compounds, such as the gases, as well as a favorable elution of methanol between gas compounds.



Pure Chromatography

Figure 1 Rxi®-624Sil MS columns resolve methyl ethyl ketone and ethyl acetate, a separation not obtained on other 624 columns.

13.496

98. n-Butylbenzene

14.579

83. trans-1,4-Dichloro-2-butene

Peaks 1. Dichlorodifluoromethane (CFC-12)	RT (min.) 2.198
2. Chloromethane	2.198
3. Vinyl chloride	2.659
4. Bromomethane	3.226
5. Chloroethane	3.434
 6. Trichlorofluoromethane (CFC-11) 7. Diethyl ether (ethyl ether) 	3.876 4.440
8. 1,1-Dichloroethene	4.909
9. 1,1,2-Trichlorotrifluoroethane (CFC-113)	4.998
10. Acetone	5.029
11. lodomethane	5.195
12. Carbon disulfide 13. Acetonitrile	5.323 5.637
14. Allyl chloride	5.715
15. Methyl acetate	5.723
16. Methylene chloride	5.981
17. <i>tert</i> -Butyl alcohol	6.234
 Acrylonitrile Methyl <i>tert</i>-butyl ether (MTBE) 	6.451 6.509
20. <i>trans</i> -1,2-Dichloroethene	6.512
21. 1,1-Dichloroethane	7.315
22. Vinyl acetate	7.359
23. Diisopropyl ether (DIPE) 24. Chloroprene	7.407 7.429
25. Ethyl <i>tert</i> -butyl ether (ETBE)	7.970
26. 2-Butanone (MEK)	8.193
27. cis-1,2-Dichloroethene	8.193
28. 2,2-Dichloropropane	8.193
29. Ethyl acetate	8.265
30. Propionitrile 31. Methyl acrylate	8.276 8.318
32. Methacrylonitrile	8.476
33. Bromochloromethane	8.507
34. Tetrahydrofuran	8.521
35. Chloroform	8.651 8.843
36. 1,1,1-Trichloroethane 37. Dibromofluoromethane	8.848
38. Carbon tetrachloride	9.026
39. 1,1-Dichloropropene	9.037
40. 1,2-Dichloroethane-d4	9.246
41. Benzene 42. 1,2-Dichloroethane	9.262 9.334
42. 1,2-Dichoroethane 43. Isopropyl acetate	9.334 9.340
44. Isobutyl alcohol	9.421
45. tert-Amyl methyl ether (TAME)	9.421
46. Fluorobenzene	9.598
47. Trichloroethene 48. 1,2-Dichloropropane	9.976 10.243
49. Methyl methacrylate	10.245
50. 1,4-Dioxane (ND)	10.299*
51. Dibromomethane	10.326
52. Propyl acetate	10.346
53. 2-Chloroethanol (ND) 54. Bromodichloromethane	10.368* 10.496
55. 2-Nitropropane	10.490
56. cis-1,3-Dichloropropene	10.904
57. 4-Methyl-2-pentanone (MIBK)	11.026
58. Toluene-D8	11.148
59. Toluene 60. <i>trans</i> -1,3-Dichloropropene	11.210 11.407
61. Ethyl methacrylate	11.435
62. 1.1.2-Trichloroethane	11.585
63. Tetrachloroethene	11.662
64. 1,3-Dichloropropane 65. 2-Hexanone	11.729 11.749
66. Butyl acetate	11.837
67. Dibromochloromethane	11.921
68. 1,2-Dibromoethane (EDB)	12.035
69. Chlorobenzene-d5 70. Chlorobenzene	12.412
70. Chlorobenzene 71. Ethylbenzene	12.440 12.507
72. 1,1,1,2-Tetrachloroethane	12.507
73. <i>m</i> -Xylene	12.612
74. <i>p</i> -Xylene	12.612
75. <i>o</i> -Xylene 76. Styrene	12.935 12.949
77. <i>n</i> -Amylacetate	12.949
78. Bromoform	13.118
79. Isopropylbenzene (cumene)	13.226
80. <i>cis</i> -1,4-Dichloro-2-butene	13.268
 81. 4-Bromofluorobenzene 82. 1,1,2,2-Tetrachloroethane 	13.385 13.456
	10.700

 trans-1,4-Dic Bromobenzer 1,2,3-Trichlor 1,2,3-Trichlor n-Propylbenz 2-Chlorotolu 1,3,5-Trimeth 4-Chlorotolu tert-Butylber Pentachloroe 1,2,4-Trimeth sec-Butylben sec-Butylben 4-Isopropyltr 1,3-Dichlorot 1,4-Dichlorot 1,4-Dichlorot 	ne opropane teene ene vibenzene traene thane ylbenzene zene seluene (p-cymene) venzene- penzene-D4	13.515 99.1 13.526 100.1 13.565 101. N 13.657 102.1 13.657 103. H 13.657 103. H 13.751 104. N 13.965 105. 1 14.007 * ND =	-Butylbenzene ,2-Dichlorobenzene ,2-Dichlorobenzene ,2,4-Trichlorobenze levachloro-1,3-butz laphthalene ,2,3-Trichlorobenze not detected; reter njection	ropropane (DBCP) ne adiene	14.579 14.635 15.252 15.407 15.935 16.040 16.196 16.396 wet needle
26 810 820	$29 \\ -= m/z 43 \\ -= m/z 72 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 19,20 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10 \\ 10 \\ $	42, 40,41 25 25 33,34 36,37 36,37 32 35 22 29,30 31 32 35 22 21 20 31 31 30 31 30 32 35 35 30 31 30 30 30 30 30 30 30 30 30 30 30 30 30	44,45 50*,51,52 59 66 46 58 62 48 56 47 54 47 54 53** 55 5 55 5	71,72 73,74 73,76 94,85 64,65 77 80 90 90 90 90 90 90 90 90 90 9	104 102 103 0 101 10.05 10.0 100 10.0 10.0 10.0 10.
Column:	Rxi®-624Sil MS, 30 m, 0.25 mm II (cat.# 13868)	GC_E\ 0, 1.40 µm	Oven	35 °C (hold 5 min.) to 60	°C at 11 °C/min_to
Sample:	8260A Surrogate Mix (cat.# 3024 8260A Internal Standard Mix (cat	.# 30241)	Oven Temp:	220 °C at 20 °C/min. (hol	
	8260B MegaMix® Calibration Mix VOA Calibration Mix #1 (ketones)	(cat.# 30006)	Carrier Gas: Flow Rate:	He, constant flow 1.0 mL/min.	
	8260B Acetate Mix (revised) (cat California Oxygenates Mix (cat.# 502.2 Calibration Mix #1 (gases) (30465)	Detector: Mode: Transfer Line	MS Scan	
Conc.: Injection	25 ppb in RO water purge and trap split (split ratio 30		Temp.: Analyzer Type:	230 °C Quadrupole	
Inj. Temp.: Purge and Trap	225°C	,	Source Temp.: Quad Temp.:	230 °C 150 °C	
Instrument: Trap Type:	OI Analytical 4660 10 Trap		Electron Energy: Solvent Delay	70 eV	
Purge: Desorb Preheat	11 min. @ 20 °C		Time: Tune Type:	1.5 min. BFB	
Temp.: Desorb:	180 °C 0.5 min. @ 190 °C		Ionization Mode: Scan Range:		
Bake: Interface	5 min. @ 210 °C		Instrument: Notes	Agilent 7890A GC & 597	5C MSD
Connection:	injection port		Other Purge and T Sample Inlet: Sample: Water	rap Conditions: 40°C 40°C	

Water

Management:

Purge 110°C, Desorb 0°C, Bake, 240°C

Not all "624s" are Equivalent

While optimizing instrument conditions can improve sample throughput, obtaining adequate resolution depends largely on column selectivity, thermal stability, and inertness. Rxi[®]-624Sil MS columns are optimized across these parameters, and therefore provide reliable separation of critical VOCs.

Lower Bleed Means Improved Sensitivity and Longer Column Lifetime

While 624 type columns generally provide good selectivity for most volatiles, they are limited by their low thermal stability. Poor thermal stability results in phase bleed that can reduce column lifetime, decrease detector sensitivity (especially ion trap mass spectrometers), and interfere with the quantification of later eluting compounds. Rxi*-624Sil MS columns have the highest thermal stability and lowest bleed among 624 type columns due to the incorporation of phenyl rings in the polymer backbone (Table I, Figure 2). The conjugated ring system of this silarylene phase provides a more rigid structure that increases thermal stability compared to nonsilarylene phases.

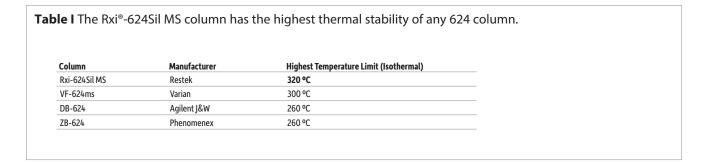
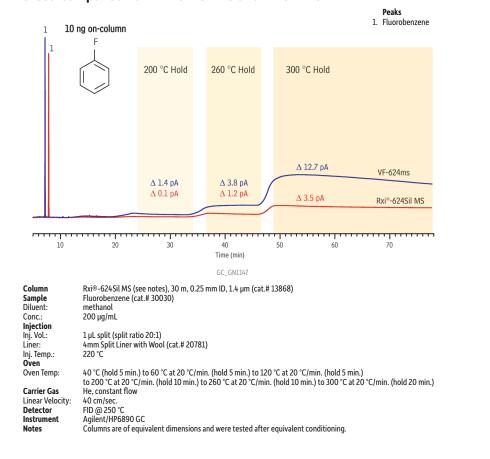


Figure 2 The Rxi[®]-624Sil MS column has the lowest bleed of any column in its class and provides true GC/MS capability.

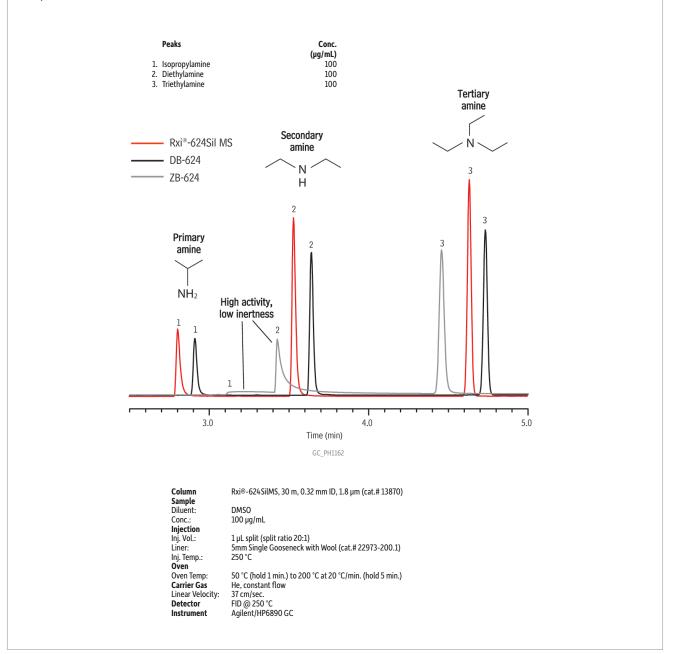


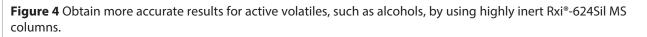
Bleed Comparison of Rxi[®]-624Sil MS and VF-624ms

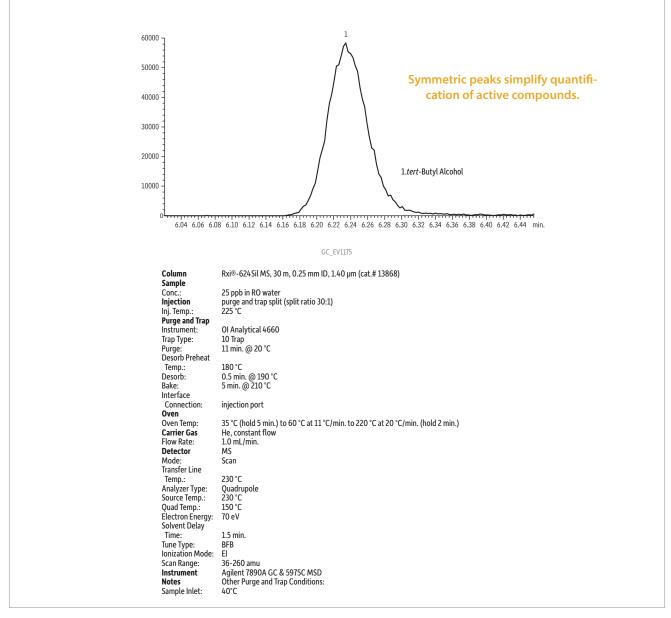
Better Peak Shape Means More Accurate Results

Rxi[®]-624Sil MS columns are the most inert 624 column available. Figure 3 shows the differences between vendor columns using primary amines, which are good indicators of column activity. The unique Rxi[®] deactivation results in symmetric peaks with minimal tailing, which improves quantitative accuracy. Minimizing tailing is especially important with concentration techniques, such as purge and trap, since the act of desorbing analytes off of the packing material results in some tailing. If a column is not inert, additional tailing due to column activity can magnify this problem. The sharp, symmetric peaks seen on Rxi[®]-624Sil MS columns allow greater resolution, higher signal-to-noise ratios, and more accurate results for active volatiles such as alcohols (Figure 4).

Figure 3 Highly inert Rxi[®]-624Sil MS columns provide better peak shape and more accurate results for active compounds.







Conclusions

Labs interested in optimizing resolution and sample throughput can adopt the conditions established here on Rxi^{*}-624Sil MS columns to maximize productivity and assure accurate, reliable results.



Questions about this or any other Restek® product? Contact us or your local Restek® representative (www.restek.com/contact-us).

Restek® patents and trademarks are the property of Restek Corporation. (See www.restek.com/Patents-Trademarks for full list.) Other trademarks in Restek® literature or on its website are the property of their respective owners. Restek® registered trademarks are registered in the U.S. and may also be registered in other countries.

© 2011 Restek Corporation. All rights reserved. Printed in the U.S.A.

www.restek.com

