

Application News

X-Ray Analysis

No. **X267**

X-Ray Fluorescence Analysis of Light Elements in Liquid Samples

- EDX-8100 and Helium Purge Unit -

Shimadzu's EDX-8100 is an energy dispersive X-ray fluorescence spectrometer with high sensitivity for light elements and can be configured with an optional helium purge unit. The helium purge unit enables high-sensitivity analysis of light elements in samples that cannot be depressurized to a vacuum state such as solutions and samples that generate gas. As one effect, detection of fluorine (F) in liquid samples is possible with the EDX-8100, whereas it was not possible with the conventional EDX-8000.

The detection limits of the elements F to K in liquid samples in a He atmosphere and differences depending on the film used to hold the samples were evaluated.

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Elements and Samples

The samples are shown in Table 1. Pure water was used to calculate the background intensity.

Table 1 Elements, Contents, and Solutions

| Element | Content [ppm] | Solution | | |
|-----------------|------------------|-------------------------------|--|--|
| F Na | 94,962 20,000 | Preparation Preparation | | |
| Mg Al | 20,000 | Preparation | | |
| | 20,000 | Preparation Atomic Absorption | | |
| Si, P, S, Cl, K | 1,000 | Standard Solution | | |
| - | - | Pure Water | | |

Sample Pretreatment

Approximately 5 mL of each sample was introduced directly into a film-covered sample container. The three films shown in Table 2 were used.

Table 2 Sample Holder Films

| Name | Thickness (µm) | Compositional Formula | | |
|---------------------|----------------|---|--|--|
| Prolene® (Chemplex) | 4 | C₃H ₆ | | |
| Polypropylene | 5 | C₃H ₆ | | |
| PET | 6 | C ₁₀ H ₈ O ₄ | | |

Detection Limits

(1) Atmosphere: Air/He

PET

Fig. 1 (next page) shows the measured spectra in air and a He atmosphere when using the polypropylene film. Table 3 shows the detection limits*1 calculated from the intensity and measurement conditions (current and integration time). The He atmosphere enabled detection of F and Na, and improved detection of Mg by 40 times, Al and Si by 8 times, and P and S by approximately 2 times.

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(2) Differences Depending on Film

Three types of films were used in measurements in a He atmosphere. Fig. 2 (next page) shows the spectra, and Table 4 shows the detection limits. Prolene (Chemplex Inc.) with a thickness of 4 μ m displayed the highest sensitivity, and the influence of the film thickness and composition on absorption increased with lighter elements.

Conclusion

Measurements of liquid samples by EDX are possible regardless of conditions such as concentration, organic or inorganic material, suspension, and viscosity.

This technique is not limited to solutions, but is also effective with samples such as the following, which are difficult or impossible to depressurize to a vacuum.

- Teeth and other biomaterials (fracture, alteration) (Care is necessary in handling, for example, considering infection.)
- Porous materials such as zeolite (time required to reach measurement vacuum)
- Fibers and clothing (porous, wet)
- · Ultrafine particles such as graphite powder (scattering)
- · Sealed and enclosed materials (bursting)

This technique is effective for quickly determining the presence and content of light elements in the abovementioned materials.

■ Measurement Conditions

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| Instrument | : EDX-8100 |
|------------------------|------------------------------|
| X-ray tube | : Rh target |
| Detector | : SDD |
| Tube voltage | : 50 [kV] |
| Tube current | : 260 to 483 [µA] (Auto) |
| Primary filter | : None (F to S) / #2 (Cl, K) |
| Collimator | : 10 [mmφ] |
| Measurement atmosphere | : Air/He |
| Integration Time | : 100 [sec] |
| Dead time | : 30 max. [%] |
| | |

Table 3 Detection Limits (1) Differences Depending on Atmosphere [ppm]

| Atmosphere | ₉ F | 11Na | 12 Mg | 13 A I | 14 Si | 15P | ₁₆ S | 17Cl | 19 K |
|--|----------------|------|--------------|---------------|--------------|-----|-----------------|------|-------------|
| Air | - | - | 3,300 | 210 | 91 | 25 | 16 | 23 | 6.3 |
| He | 41,000 | 680 | 84 | 26 | 11 | 11 | 7.8 | 15 | 5.1 |
| Table 4 Detection Limits (2) Differences Depending on Film [ppm] | | | | | | | | | |
| Film | ₉ F | 11Na | 12Mg | 13 A I | 14Si | 15P | ₁₆ S | 17CI | 19 K |
| Prolene® (Chemplex) | 18,000 | 480 | 72 | 22 | 10 | 9.7 | 7.5 | 14 | 4.7 |
| Polypropylene | 41,000 | 680 | 84 | 26 | 11 | 11 | 7.9 | 15 | 5.1 |

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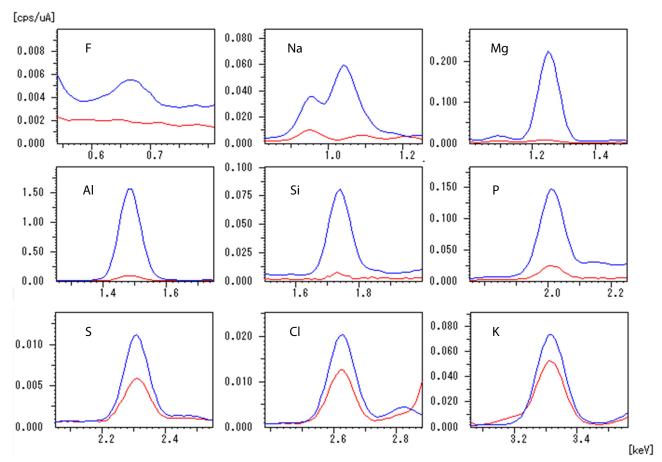


Fig. 1 X-Ray Fluorescence Spectra for F to K [Blue: He Atmosphere, Red: Air]

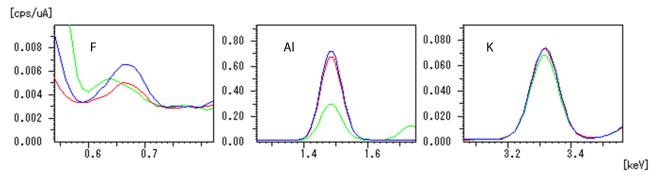


Fig. 2 X-Ray Fluorescence Spectra for F, AI, and K [Blue: Prolene (Chemplex), Red: Polypropylene, Green: PET]

*1 Calculation of detection limit Detection limit Detection limit =
$$3 \times \frac{\text{Content}}{\text{X-ray fluorescence intensity}} \times \frac{\sqrt{\text{Background intensity} \times \text{Integration time} \times \text{Current}}}{\text{Integration time} \times \text{Current}}$$
(Example: Case of Table 3, P in He atmosphere) = $3 \times \frac{1,000_{[ppm]}}{1.217437_{\left[\frac{CPS}{\mu A}\right]}} \times \frac{\sqrt{0.536691_{\left[\frac{CPS}{\mu A}\right]} \times 100_{\left[\text{sec}\right]} \times 283_{\left[\mu A\right]}}}{100_{\left[\text{sec}\right]} \times 283_{\left[\mu A\right]}} = 10.73_{\left[ppm\right]} = 11_{\left[ppm\right]}$

 $\label{prolene} Prolene\ is\ a\ registered\ trademark\ of\ Chemplex\ Industries\ Inc.$

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