GC Method Developement

What to Consider

The Sample

Method of injection

Inlet

Detector

Carrier Gas

Column

COMPOUND REQUIREMENTS FOR GC

Only 10-20% of all compounds are suitable for GC analysis

The compounds must have:

- ✓ Sufficient volatility
- ✓ Thermal stability

NO Inorganic Acids and Bases

Be mindful of salts!

Sample Considerations

1. Sample matrix
   - residues?
   - dirty samples?
2. Analyte Composition
   1. Isomers?
   2. Polar vs. non-Polar?
   3. Organic Acids?
   4. Light Gases?
   5. Nobel Gases?
   6. Halogens?

Sample Residues

Semi-volatile residues

- Bake out
- Back flush

Non-volatile residues

- Guard column
- Back flush

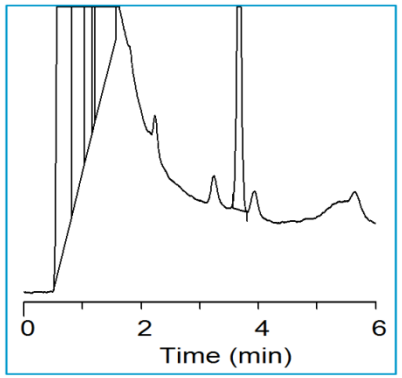
Dirty Samples

- Sample clean up?
- Back flush

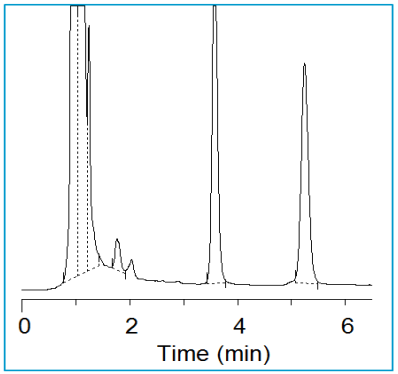
Perform Sample Preparation

- To acquire desired sensitivity/selectivity
- To reduce contamination/carryover issues
- Increase inlet maintenance interval
- Use of sensitive and expensive instruments: *Protect your investment!!!*

Pesticides in Avocado *without* SP



Pesticides in Avocado with SP



We have thought about the sample
…What's next?

Let’s Get the Sample Onto the Column…

Manual Injection

Liquid Injection

Headspace

Purge & Trap

Gas Sampling Valve

SPME

Thermal Desorption

Custom

The Inlet

Volatiles Interface

Cool-On-Column

Purged Packed

PTV/MMI

Split / Splitless

Multi-Mode

Cool-On-Column

* Good for Labile Samples

    Sample is deposited “ON” the column

    Temperature of inlet follows Oven Temperature

- Good for ‘Active’ analytes
    - Minimizes inlet discrimination
    - No inlet Liner*
- Good for Trace Analysis
- Guard Column Highly Recommended

Purged Packed

Good for HIGH flow applications

Used with Packed columns

Can be used with 0.53 mm and 0.32 mm ID columns

**Has a minimal capacity for sample expansion

Back Flash

Split / Splitless

Mode	Sample Concentration	Sample to Column	Comments
Split	High	Very Little	
Pulsed Split	High	Very Little	Useful with large injections
Splitless	Low	All	
Pulsed Splitless	Low	All	Useful with large injections. *better transfer of sample to column*

SPLIT INJECTOR

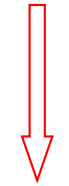
Split Ratio

- Too low: Poor peak shape
  -Column overload
- Too high: Poor sensitivity
  -Wastes carrier gas (gas saver)
- Usually non-linear
  -Do not use ratio as a dilution factor

Minimum Recommended Split Ratio

mm I.D.	Lowest ratio
0.10	1:50 - 1:75
0.18 - 0.25	1:10 - 1:20
0.32	1:8 - 1:15
0.53	1:2 - 1:5

Higher flow rates



Want to have 20 mL/min flow through the inlet

Multimode

Mode	Sample Concentration	Sample to Column	Discussion
Split	High	Low	
Pulsed Split	High	Low	
Splitless	Low	All	
Pulsed Splitlss	Low	All	
Solvent Vent	Low	All	Multiple Injections concentrate sample and vent solvent
Direct	Low	All	

Sample Expansion…Liners?

Split / Splitless Inlet

Multimode Inlet

Use the same liners

Packed inlet

PTV

Inlet Liners - Purpose

Glass Inlet Liners provide an “inert” space for liquid samples to be uniformly vaporized to a gas and moved to the column.

Liquid-gas phase change involves a significant change in volume.

Gaseous sample volume depends on

- the solvent type
- column head pressure
- temperature of inlet

These aspects should be optimized for your sample volume and application.

Solvent (1µL, ambient)	Volume (µL at 250°C and 20psig)
n-Hexane	140
Acetone	245
Acetonitrile	350
Methanol	450
Water	1010

See “A Practical Guide to the Care, Maintenance, and Troubleshooting of Capillary GC Systems”, Third Revised Edition, by Dean Rood, Wiley-VCH, New York, 2001.

Liners - 3 Key Aspects Govern Applications

Liner Volume

Liner Treatments or Deactivation

Special Characteristics (glass wool, cup, taper, etc.)

When choosing a liner for your application, consider all three aspects to give you the best chromatography.

You must also determine what type of inlet is in your GC

Then consider the application itself, and the types of liners and injection techniques used for it:

- Split
- Splitless

Liner Volume

Choose a liner with enough volume to accommodate the vaporized sample.

Important, especially for polar solvents with large vapor volumes.

If vapor volume of sample exceeds liner volume, samples may back up (backflash) into carrier gas supply lines, causing ghost peaks and reproducibility problems in chromatography.

Liner Volume (contd.)

Agilent liners are primarily 2mm or 4mm in inner diameter (without tapers and additional features) and 78mm long.

- Thus, 2mm liners hold approx. 0.245 mL or 245 μL of vapor

  4mm liners hold approx. 0.972 mL or 972 μL of vapor

Recommended injection volumes are 1-2μL or less for organic solvents, 0.5μL for water.

Liner Volume

How Do we Calculate the Vapor Volume?

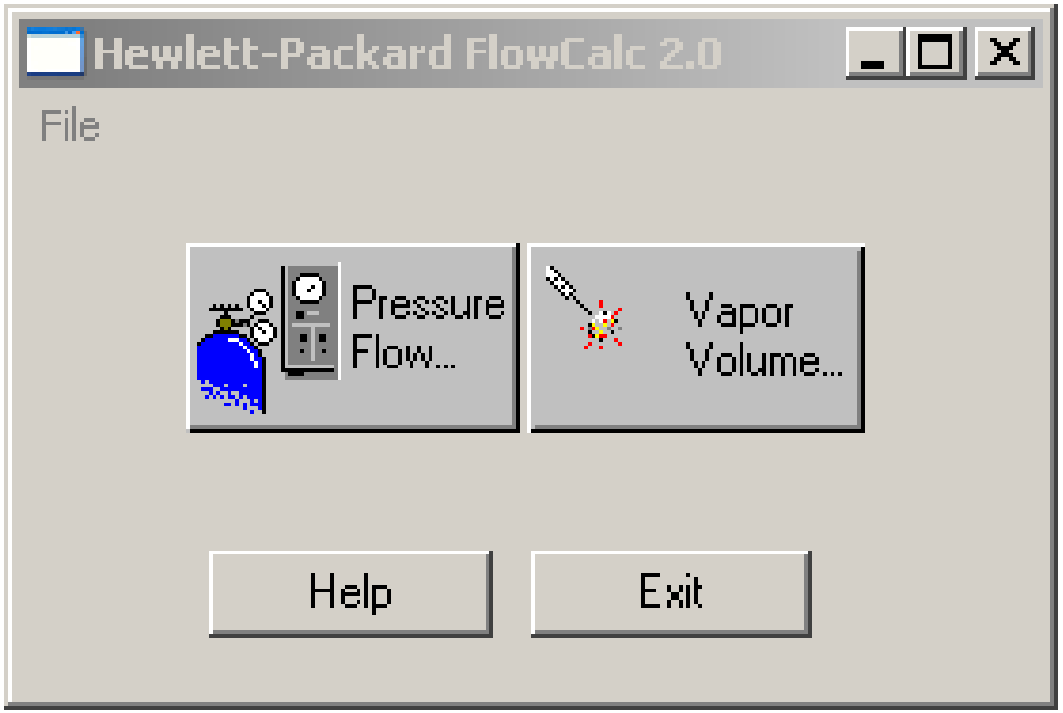
Pressure / Flow Calculator

Free download from our Website

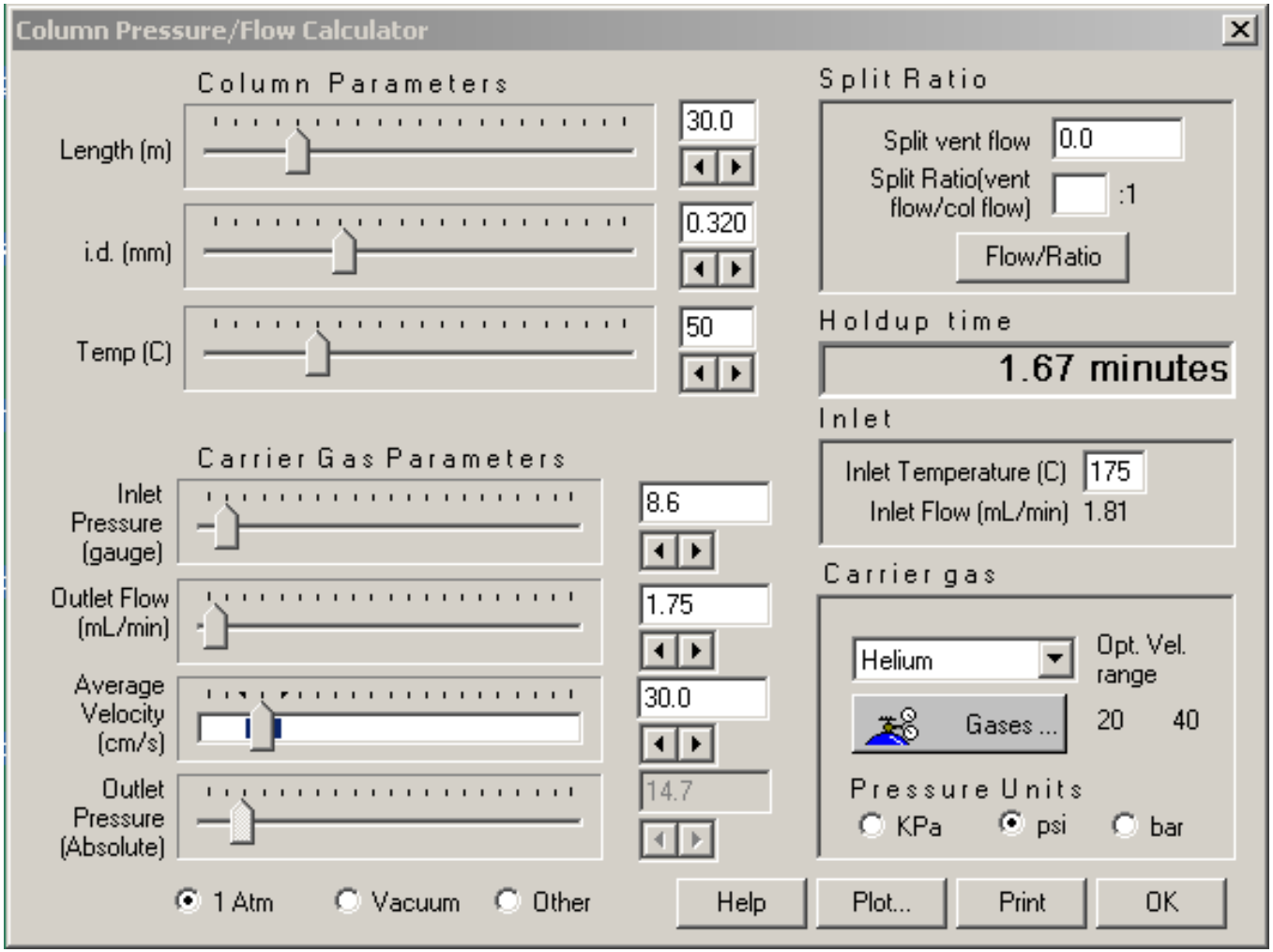
www.chem.agilent.com

https://www.agilent.com/en-us/support/gas-chromatography/gccalculators

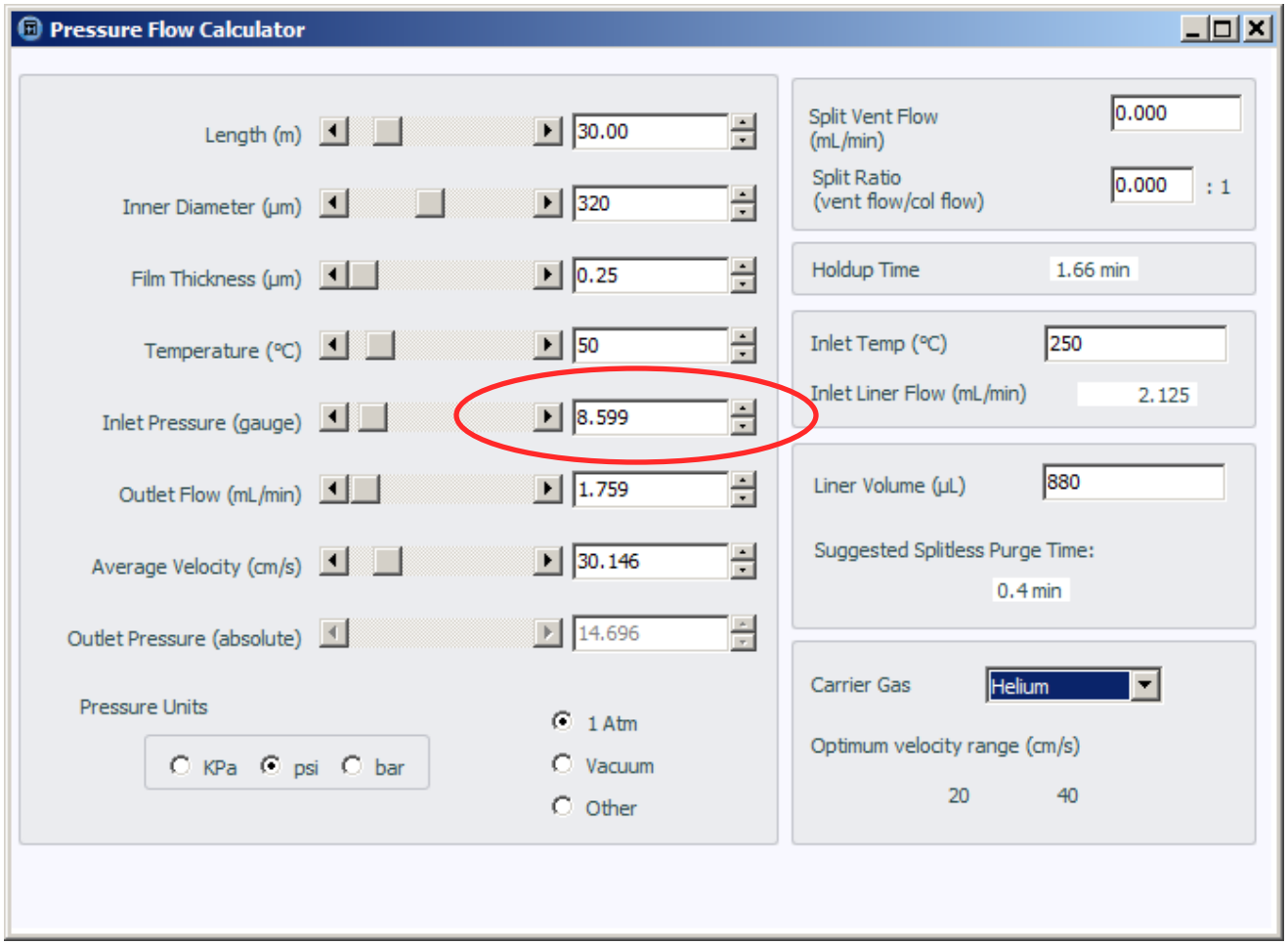
Pressure / Flow Calculator



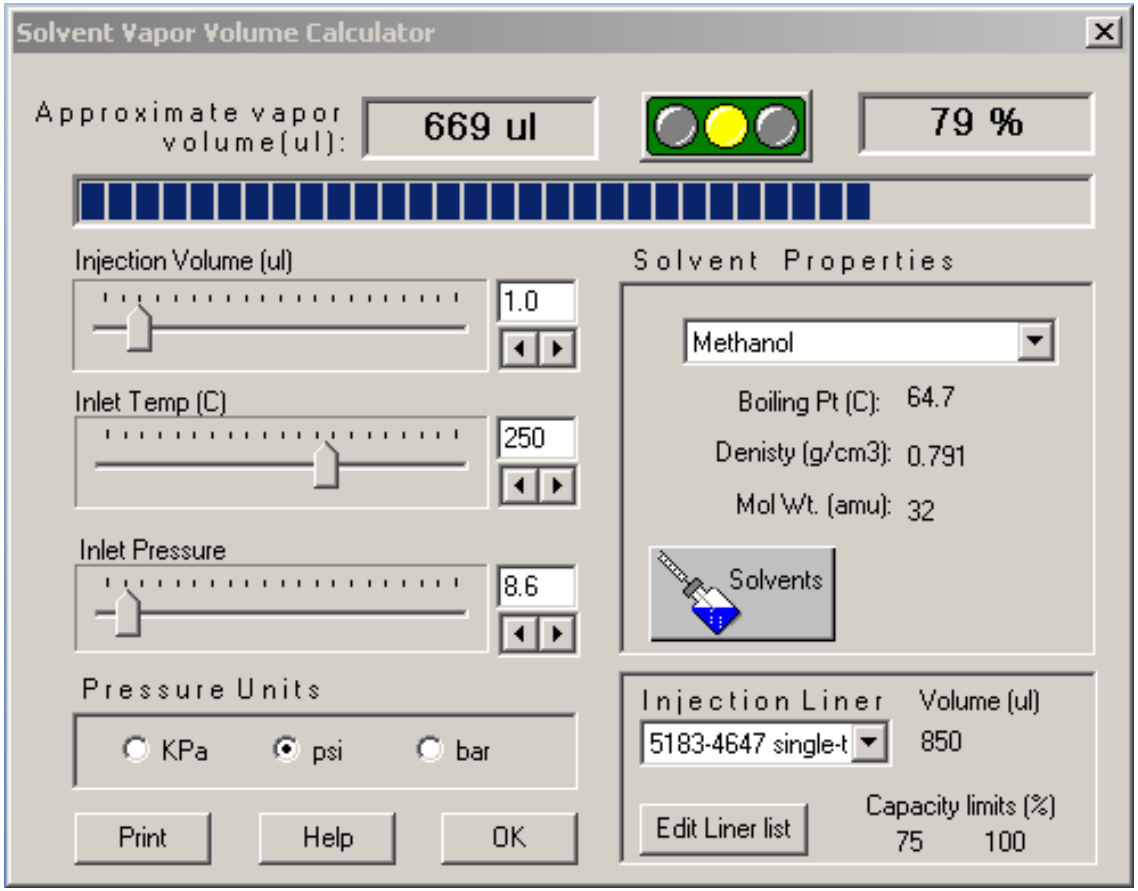
Determine what the inlet pressure will be:



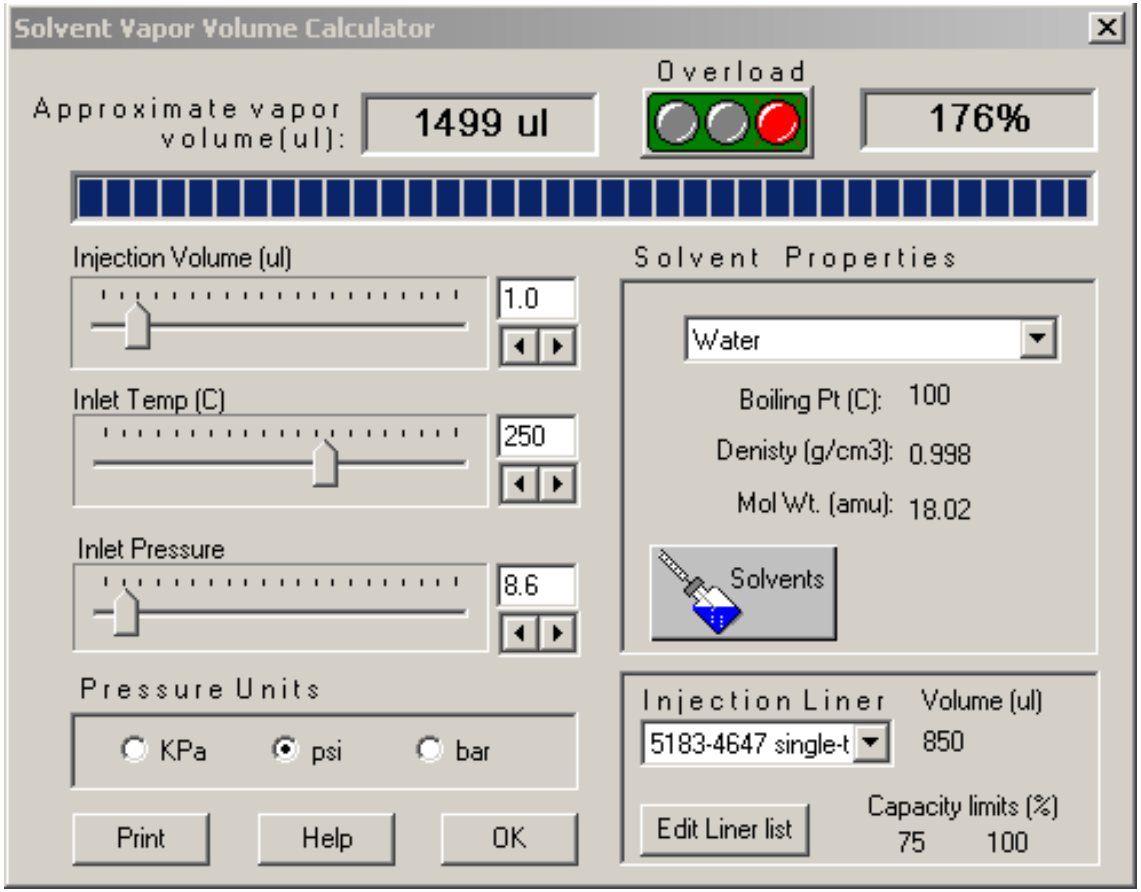
Determine what the inlet pressure will be:



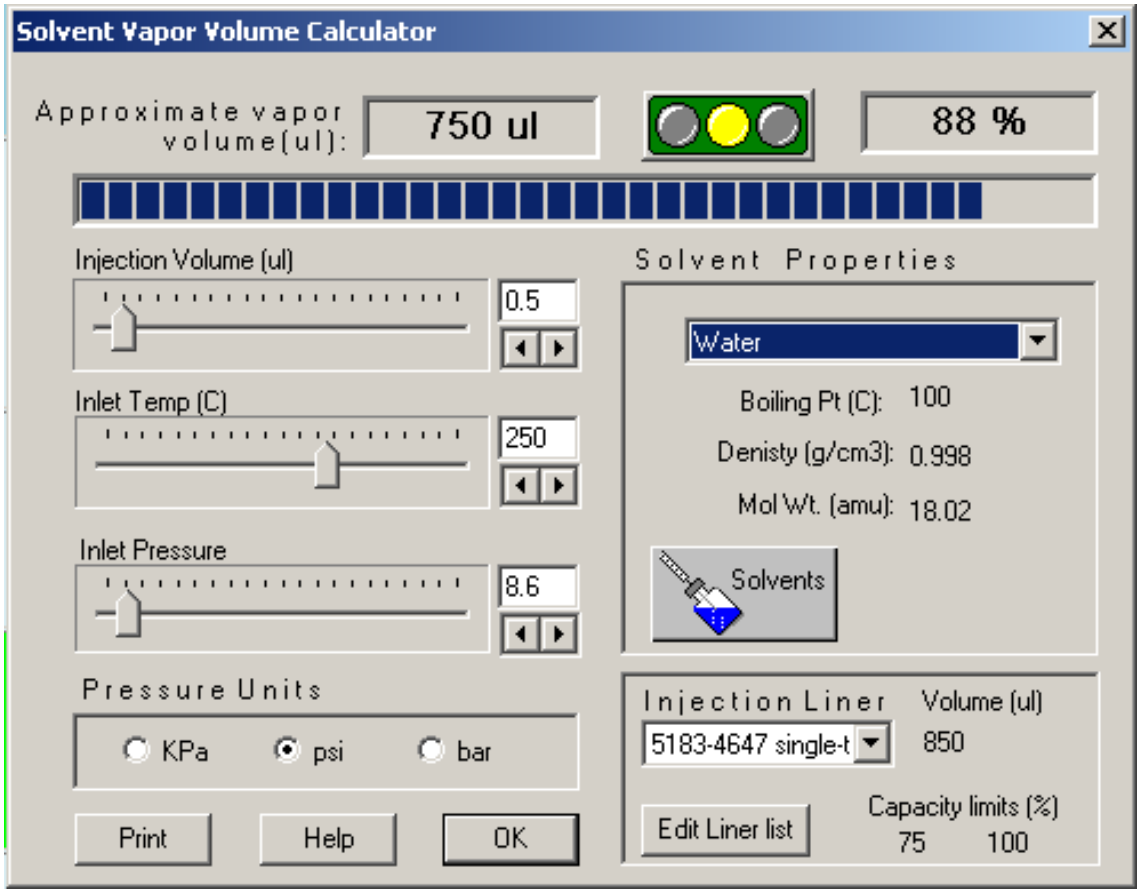
Test Inlet Conditions For Solvent Expansion



Water as Solvent

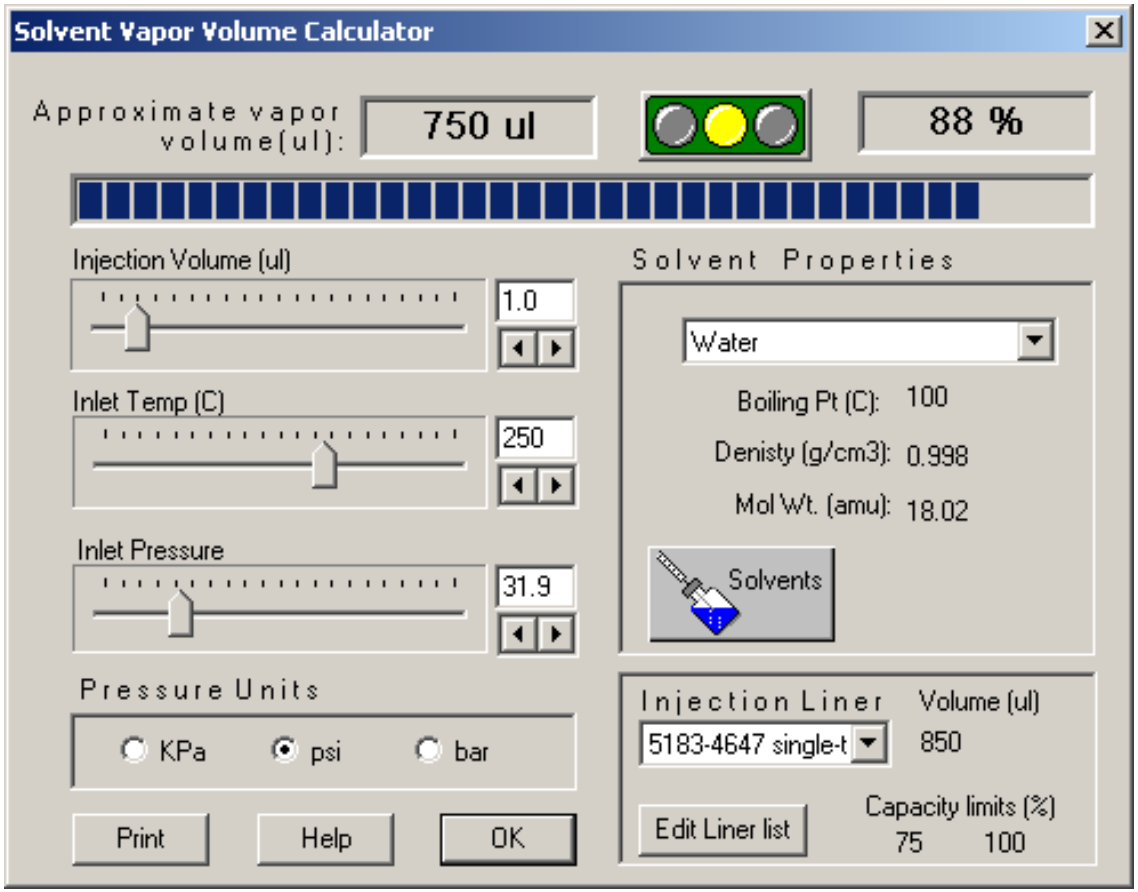


Water as Solvent
Cut Injection Volume in Half



Water as Solvent

Pulsed Injection



Liner Treatments or Deactivation

Minimizes possibility of active sample components from adsorbing on active sites on the liner or glass wool surface.

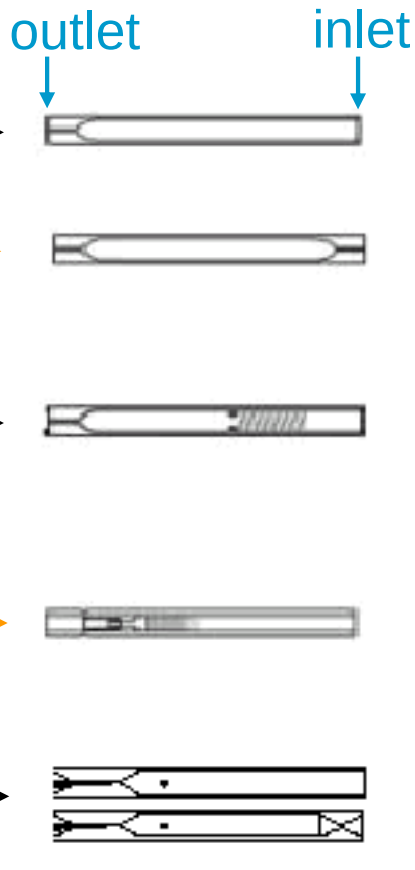
Unwanted sample adsorption leads to tailing peaks and loss of response for polar compounds.

Although not necessary for all applications, deactivated liners provide added insurance against possible sample adsorption.

Deactivation of borosilicate glass liners is often done with a silylating reagent like Dimethyldichlorosilane (DMDCS)

Special Characteristics

Some liners have special features that are necessary for different injection techniques. For example:



Taper (gooseneck), minimizes sample contact with gold seal.

Dual taper, also minimizes sample contact with inlet weldment and reduces potential for backflash.

Glass wool and shelf to hold it in place, prevents non-volatiles from reaching column and removes residual sample from needle. Glass wool should be deactivated.

Jennings cup, normally used for efficient sample mixing in split inlets, reduces sample discrimination and prevents non-volatiles from reaching the column. Not for very dirty samples.

Press fit (direct) connection end to hold capillary column firmly (virtually all sample goes onto the column). Side hole needed for Electronic Pressure Control with direct connect liners.

Special Characteristics (contd.)

Other special characteristics include:

- Baffles
- Spiral paths
- Glass or ceramic frits or beads
- Laminar cups (elongated version of Jennings cups)
- Column packings with stationary phases

All designed to provide:

- a turbulent sample flow path for sample mixing
- protrusions, barriers, or adsorbents to collect high molecular weight sample components or particles
- surfaces for efficient vaporization of sample components.

Split Injection Liners

Liner	Part No.	Comments
	5190-2294	Simplest split liner, glass wool, UI deactivation, large volume, 990µL volume. Use for general purpose. Also used for Splitless mode.
 Glass nub	5190-2295	Glass wool (held near needle entrance to remove residual sample on needle), deactivated, 870µL volume. Glass nub ensures that gap remains below liner for split injection. Efficient, for most applications, including active compounds. Fail-safe insertion into injection port. Needle length is important.
	18740-80190	Liner with Jennings cup, no glass wool, 800µL volume. Use for general purpose applications, high and low MW compounds. Reduces inlet discrimination.
	18740-60840	Liner with Jennings cup, glass wool, and column packing, 800µL volume. For dirty samples, traps non-volatiles and particulates well. For high and low MW compounds. Not recommended for use with EPC.

Splitless Injection Liners

Liner	Part No.	Comments
	5190-2292	Single taper, deactivated, 900µL volume. Taper isolates sample from metal seal, reducing breakdown of compounds that are active with metals. For trace samples, general application.
	5190-2293	Single taper, deactivated, with glass wool, 900µL volume. Glass wool aides volatilization and protects column. For trace (dirty) samples.
	5190-3983	Double taper, deactivated, 800µL volume. Taper on inlet reduces chance for backflash into carrier gas lines. High efficiency liner for trace, active samples.
	G1544-80730 G1544-80700	Direct connect liners, single and dual taper, deactivated. Capillary column press fits into liner end, eliminating sample exposure to inlet. Ultimate protection for trace, active samples. Side hole permits use with EPC.

GLASS WOOL
Liner Packing Recommendations

Amount, size and placement must be consistent for consistent results

Can be broken upon installation into the liner, exposing active sites

Liner deactivation with glass wool plug in place is ideal

GLASS WOOL
Placement in Liner

Near top of liner:

- Wipes syringe needle of sample
- Can improve injector precision
- Helps to prevent backflash

Near bottom of liner:

- Helps in volatilization of high MW components
- Increases mixing

Both positions help retain *some* non-volatile residues from reaching the column

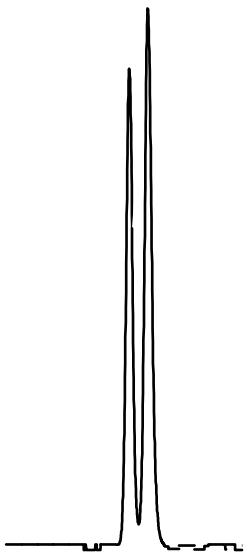
Carrier Gas Considerations

- Carries the solutes down the column
- Selection and velocity influences efficiency and retention time

RESOLUTION VS. LINEAR VELOCITY

Helium

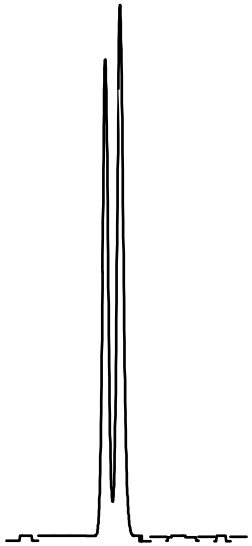
Resolution of 1.5 = baseline resolution



R = 1.46

30 cm/sec

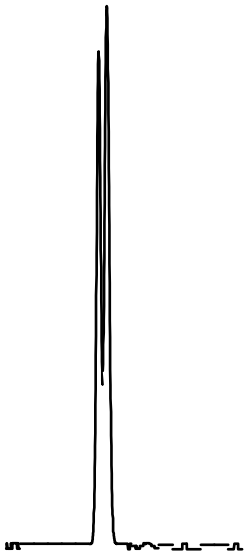
4.4 psig



R = 1.31

35 cm/sec

5.1 psig



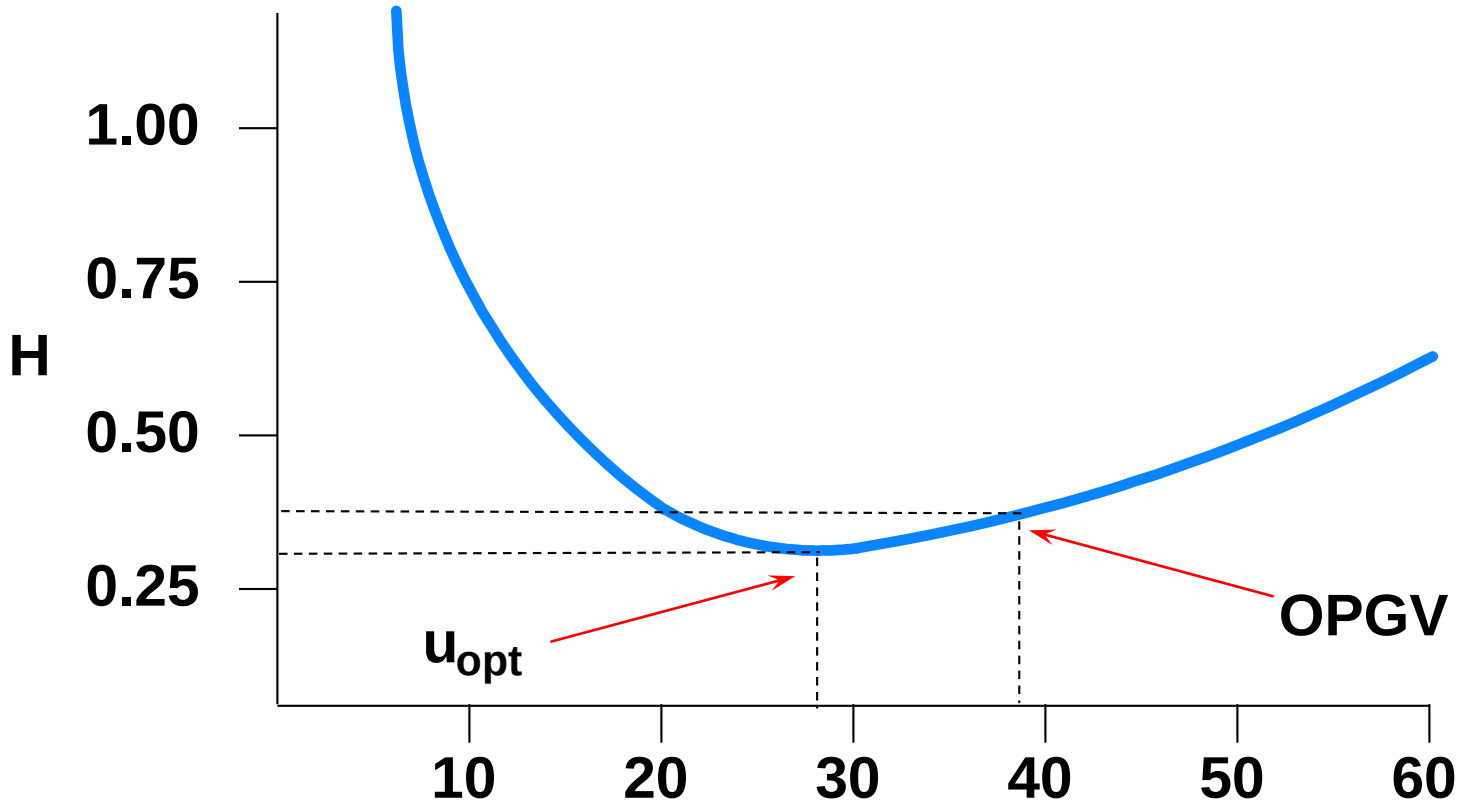
R = 0.97

40 cm/sec

5.8 psig

DB-1, 15 m x 0.32 mm ID, 0.25 um
60°C isothermal
1,3- and 1,4-Dichlorobenzene

VAN DEEMTER CURVE



$\overline{u}_{opt}$ and OPGV

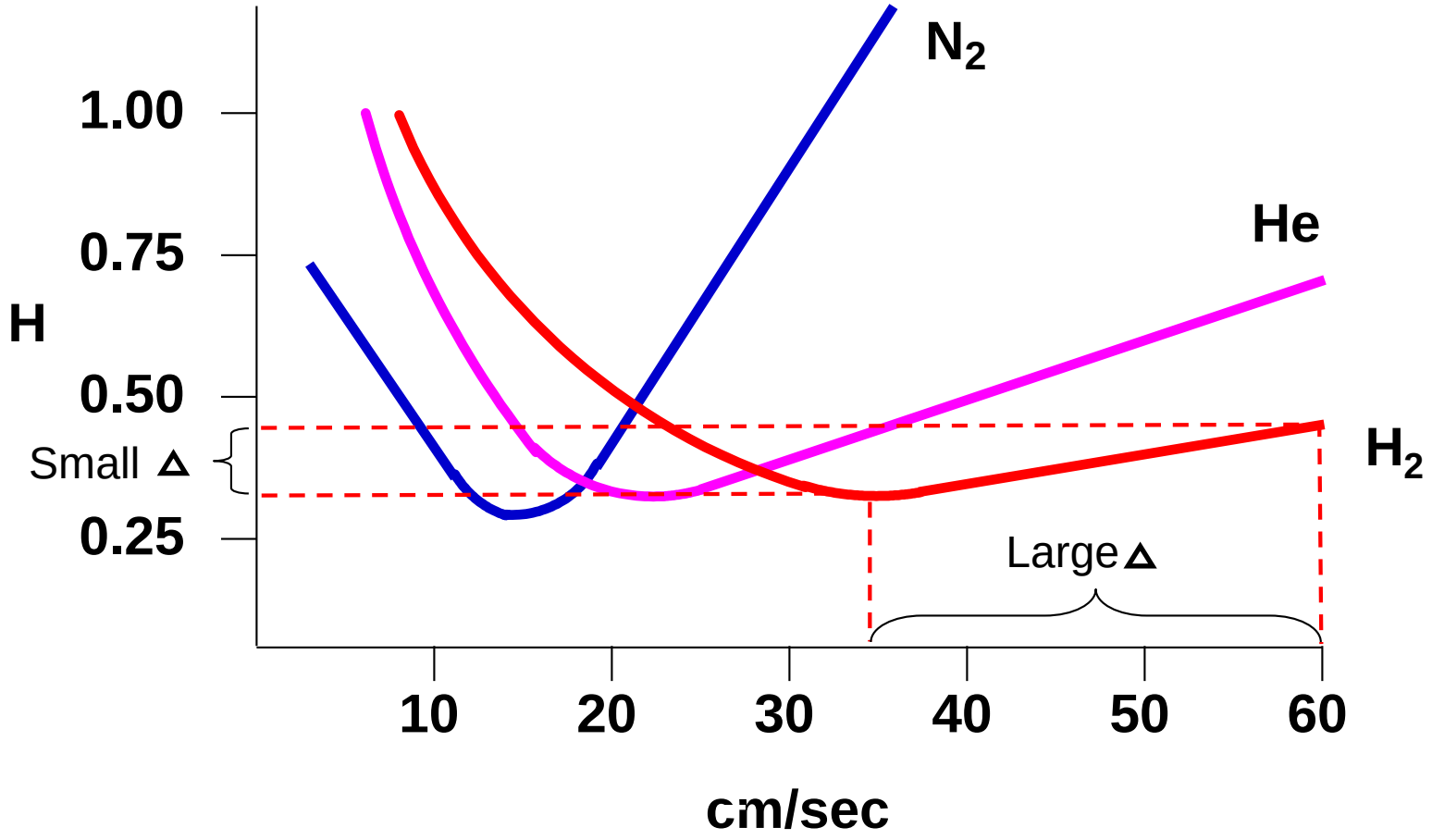
$\overline{u}_{opt}$: Maximum efficiency

OPGV: Optimal practical gas velocity

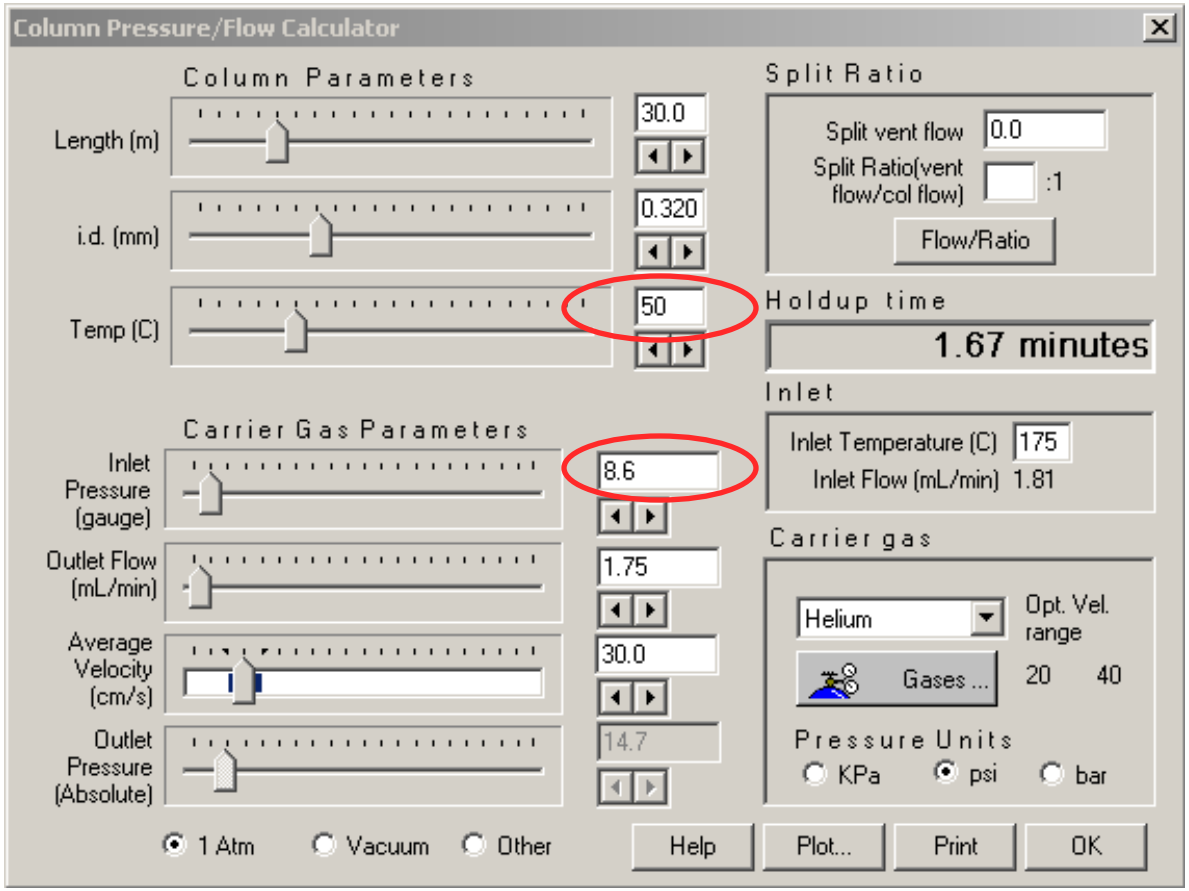
Maximum efficiency per unit time

$1.5 – 2 \times \overline{u}_{opt}$

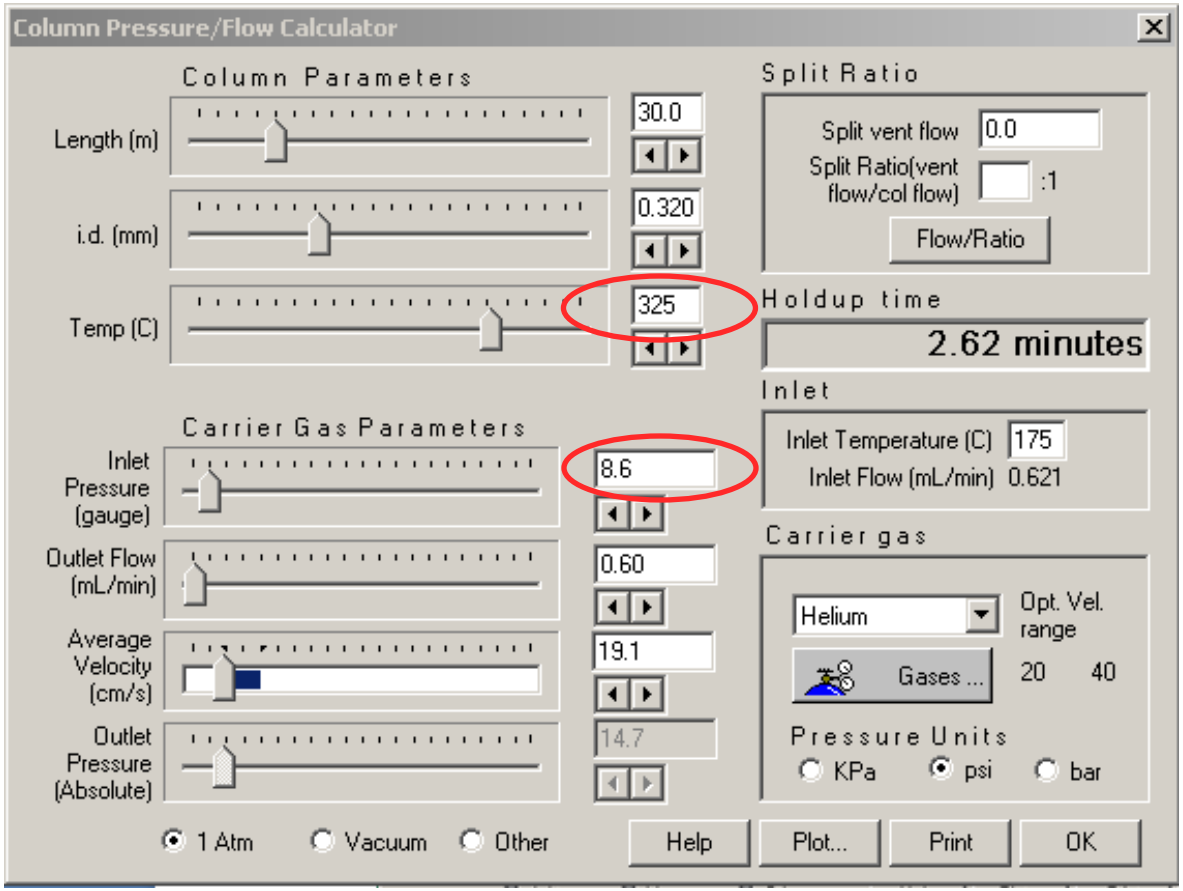
VAN DEEMTER CURVES



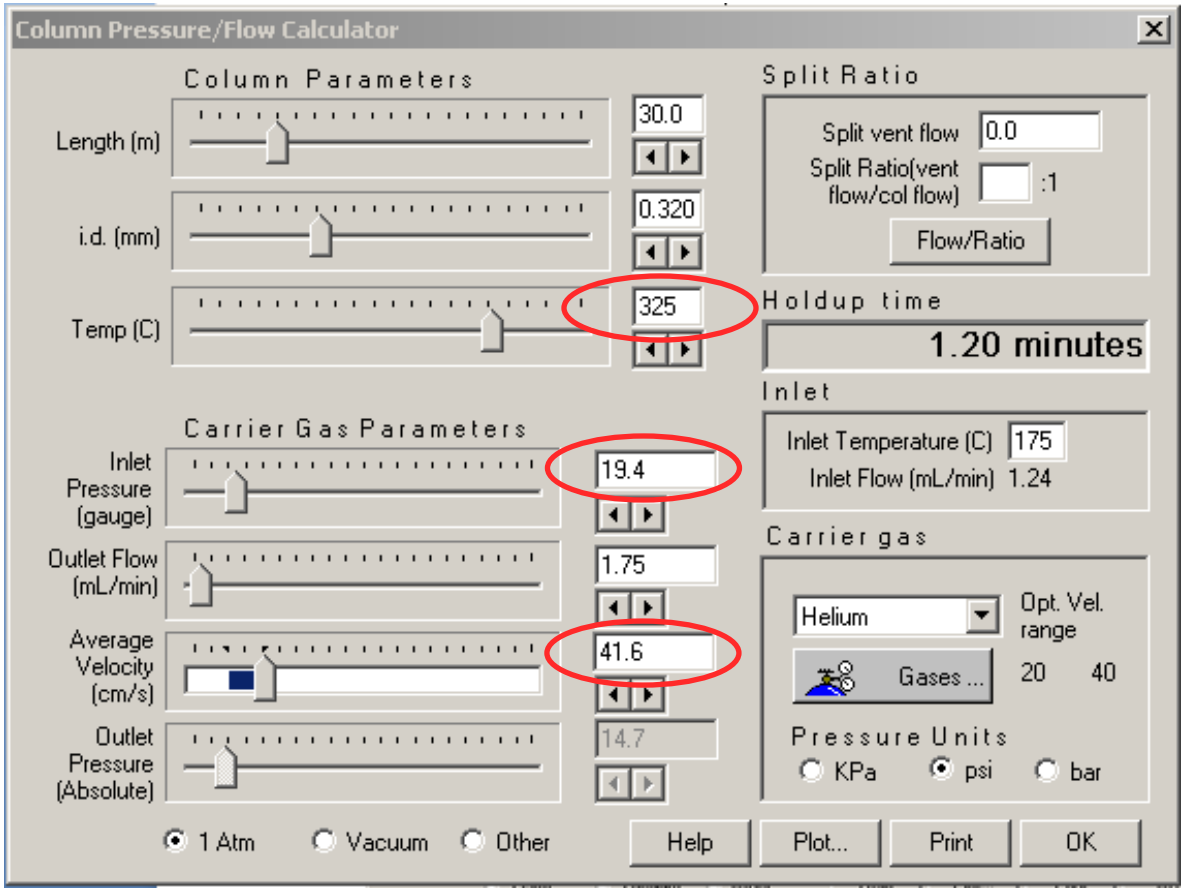
What Happens to the Flow as Oven Temp Increases?



Carrier Gas: Constant Pressure



Carrier Gas: Constant Flow



Detectors

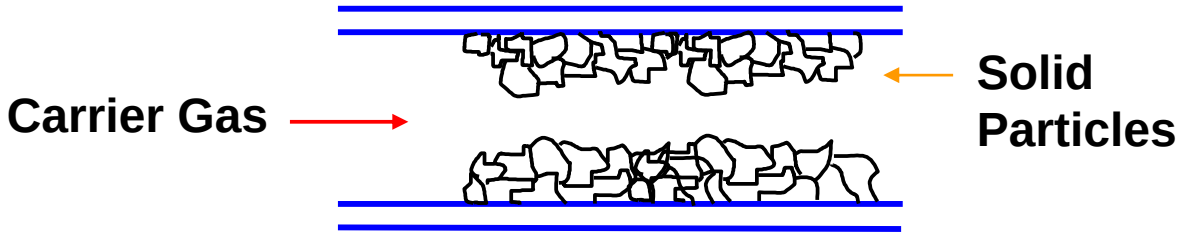
Detector	Dynamic Range		MDL
TCD	10^5	Universal	400 pg Tridecane
FID	10^7	Responds to C-H bonds	1.8 pg Tridecane
ECD	$5x10^5$	Responds to free electrons	6 fg/mL Lindane
NPD	10^5	Specific to N or P	0.4 pgN/s 0.06 pg P /s
FPD	10^3S, 10^4P	Specific to S or P	60 fg P/s 3.6 pg S/s
SCD	10^4	Specific & Selective to S	0.5 pg S/s
NCD	10^4	Specific & Selective to N	3 pg N/s
MSD		Universal	S/N 400:1 1 pg/uL OFN

Selecting the RIGHT Column

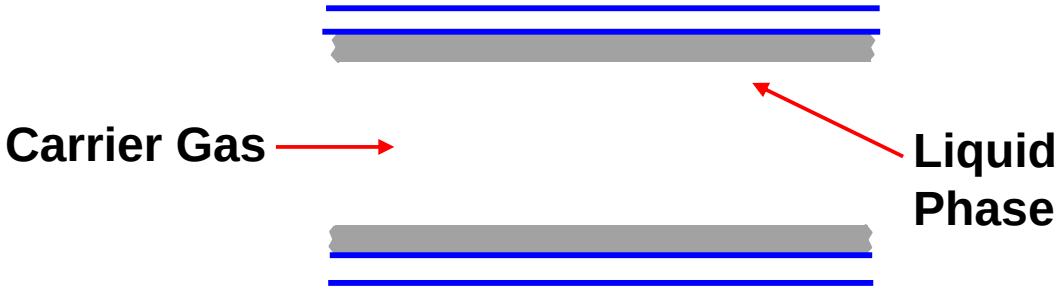
Understanding the Stationary Phase

CAPILLARY COLUMN TYPES

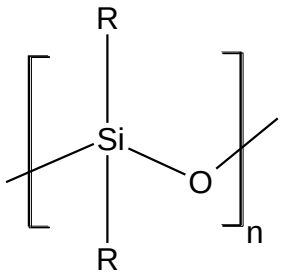
Porous Layer Open Tube (PLOT)



Wall Coated Open Tube (WCOT)

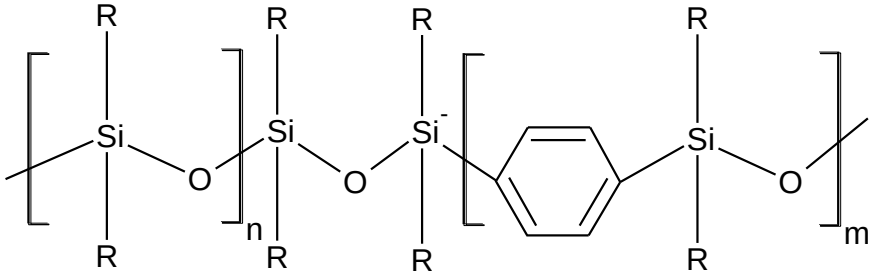


STATIONARY PHASE POLYMERS

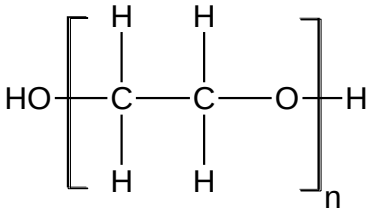


Siloxane

R= methyl, cyanopropyl, cyanopropylphenyl, trifluoropropyl



Arylene



Polyethylene glycol backbone

Selectivity Interactions

- Dispersion
- Dipole
- Hydrogen bonding

Selectivity
Interaction Strengths

Phase	Dispersion	Dipole	H Bonding
Methyl	Strong	None	None
Phenyl	Strong	None	Weak
Cyanopropyl	Strong	Strong	Moderate
Trifluoropropyl	Strong	Moderate	Weak
PEG	Strong	Strong	Moderate

Selecting the Correct Column

Match analyte polarity to column polarity
‘Like dissolves like’

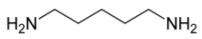
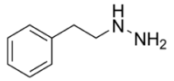
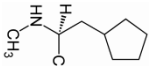
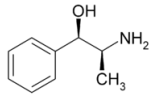
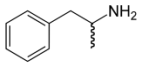
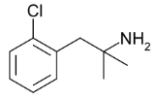
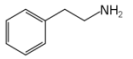
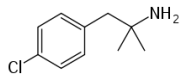
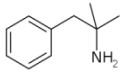
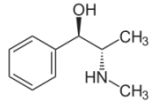
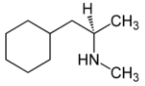
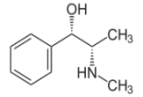
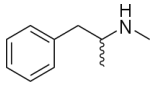
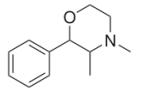
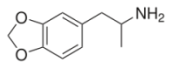
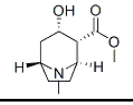
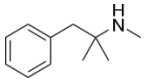
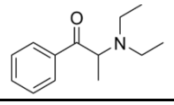
Look for unique interactions that analytes may have with a phase

Use preexisting information

Use the Agilent GC Application Support Team:
gc-column-support@agilent.com

Now Let’s Apply What We Have Learned

Sample List (drugs)

1. Cadaverine		11. Phenelzine	
2. Cyclopentamine		12. Phenylpropanolamine	
3. Amphetamine		13. Clortermine	
4. Phenethylamine		14. Chlorphentermine	
5. Phentermine		15. Ephedrine	
6. Propylhexedrine		16. Pseudoephedrine	
7. Methamphetamine		17. Phendimetrazine	
8. Methenamine		18. MDA	
9. Amantidine		19. Ecgonine methyl ester	
10. Mephentermine		20. diethylpropion	

Starting Method Parameters

Column: DB-5 30m X 0.32mm X 0.25um

S/Sl Inlet: Split 50:1 Temp 250^{o}

FID: Temp 350^{o}

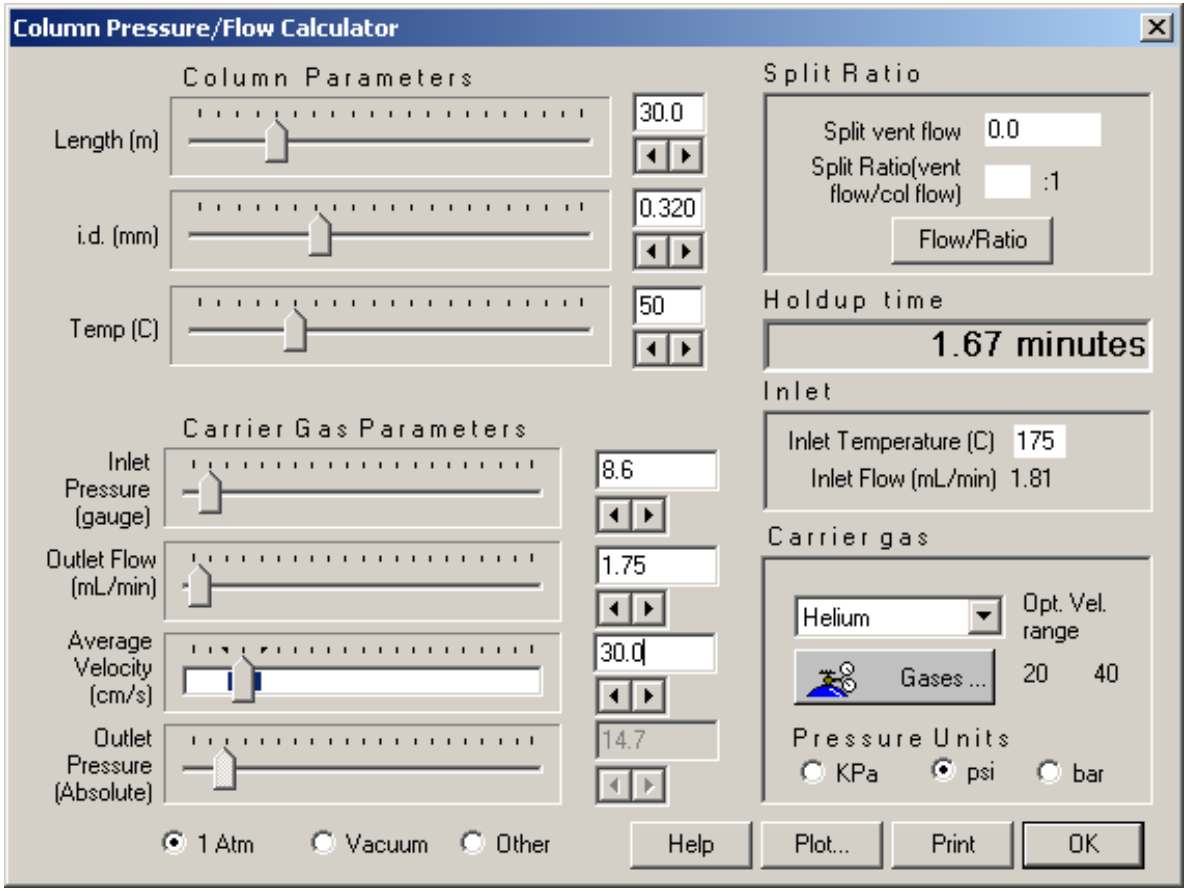
Carrier: He

Constant flow 30 cm/sec

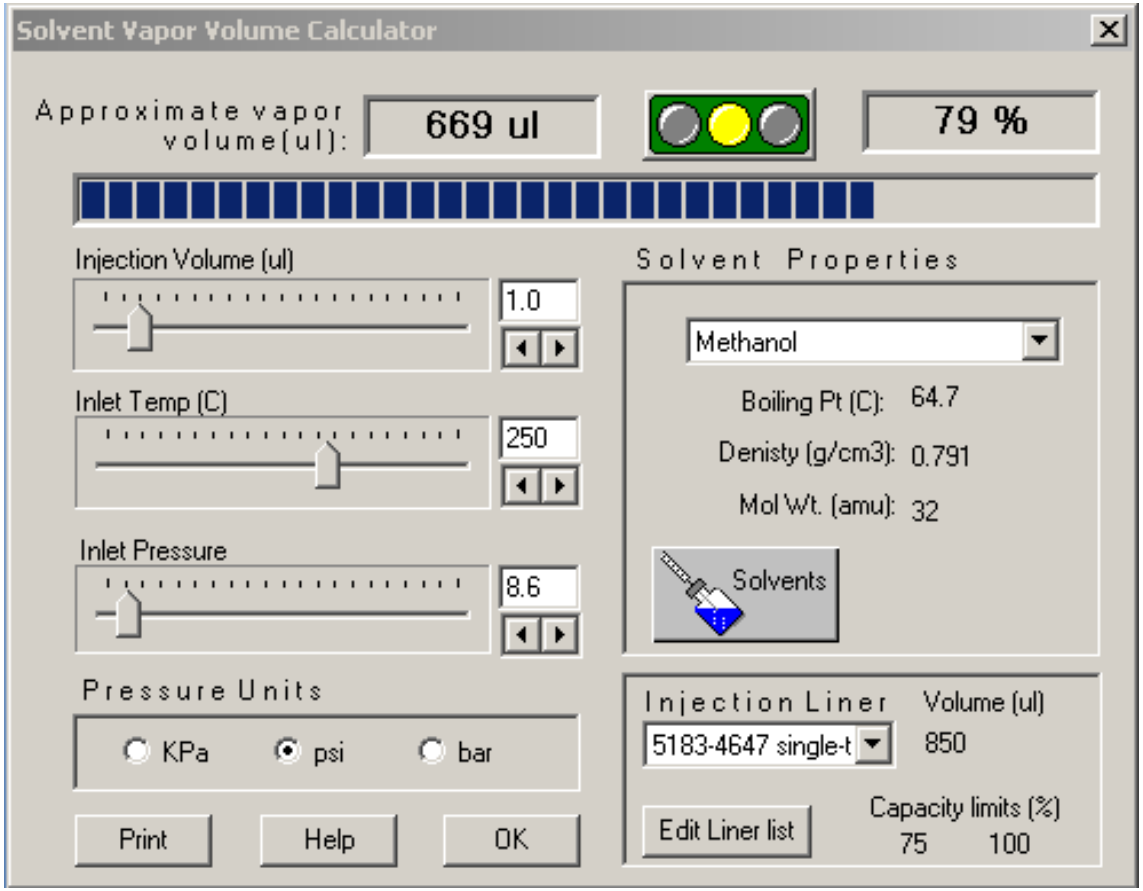
Oven: 50^{o}C Hold for 5 min

10^{o}C/min to 325^{o}C Hold for 5 min

Am I Going to Have Backflash?

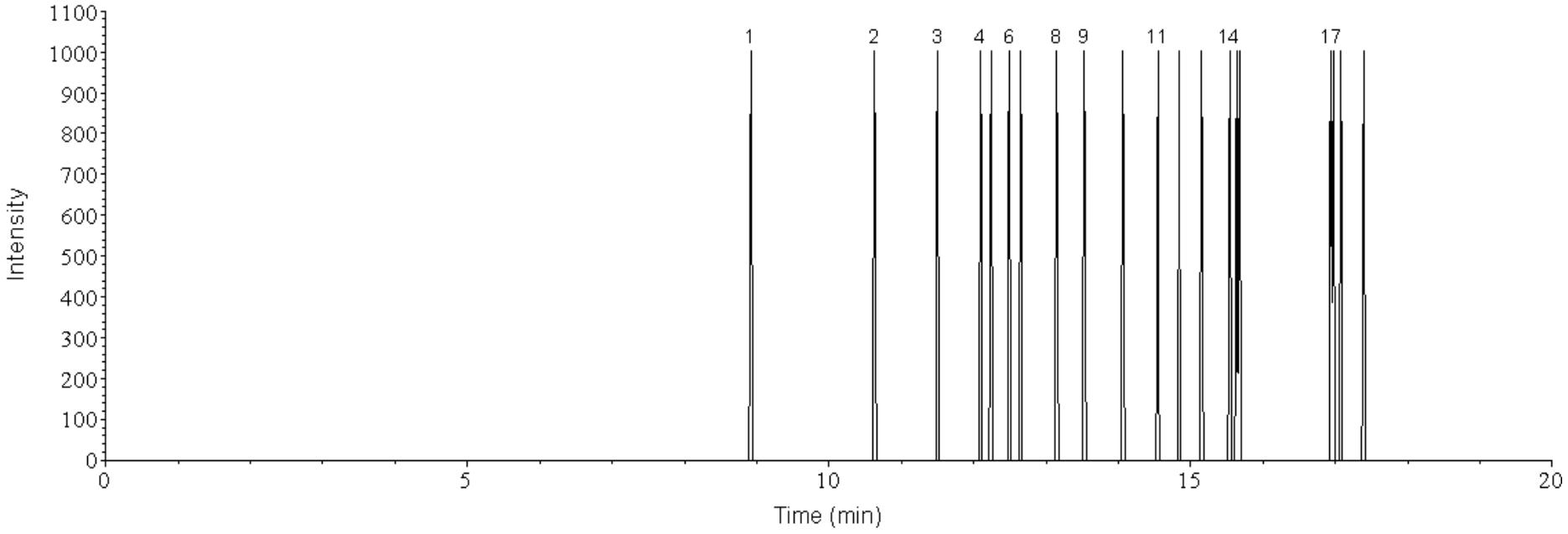


Injection Volume / Solvent Expansion

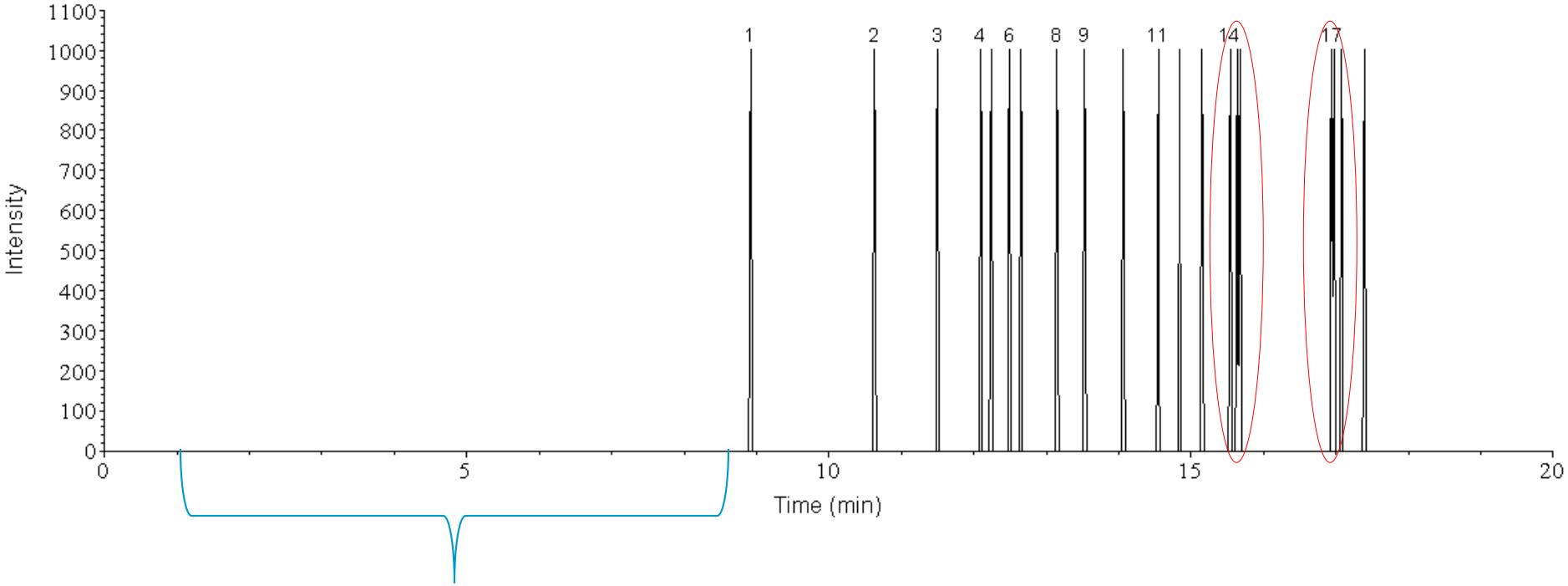


Developing Temperature Program
Initial Run

Initial Temp: 50°C Hold for 5 min
Ramp 10°C/min to 325°C Hold for 5 min



Developing Temperature Program
Initial Run - Define Areas for Improvement



Next Step…

When does the first peak come out?

~9 minutes

What temperature does it come out at?

Temp program:

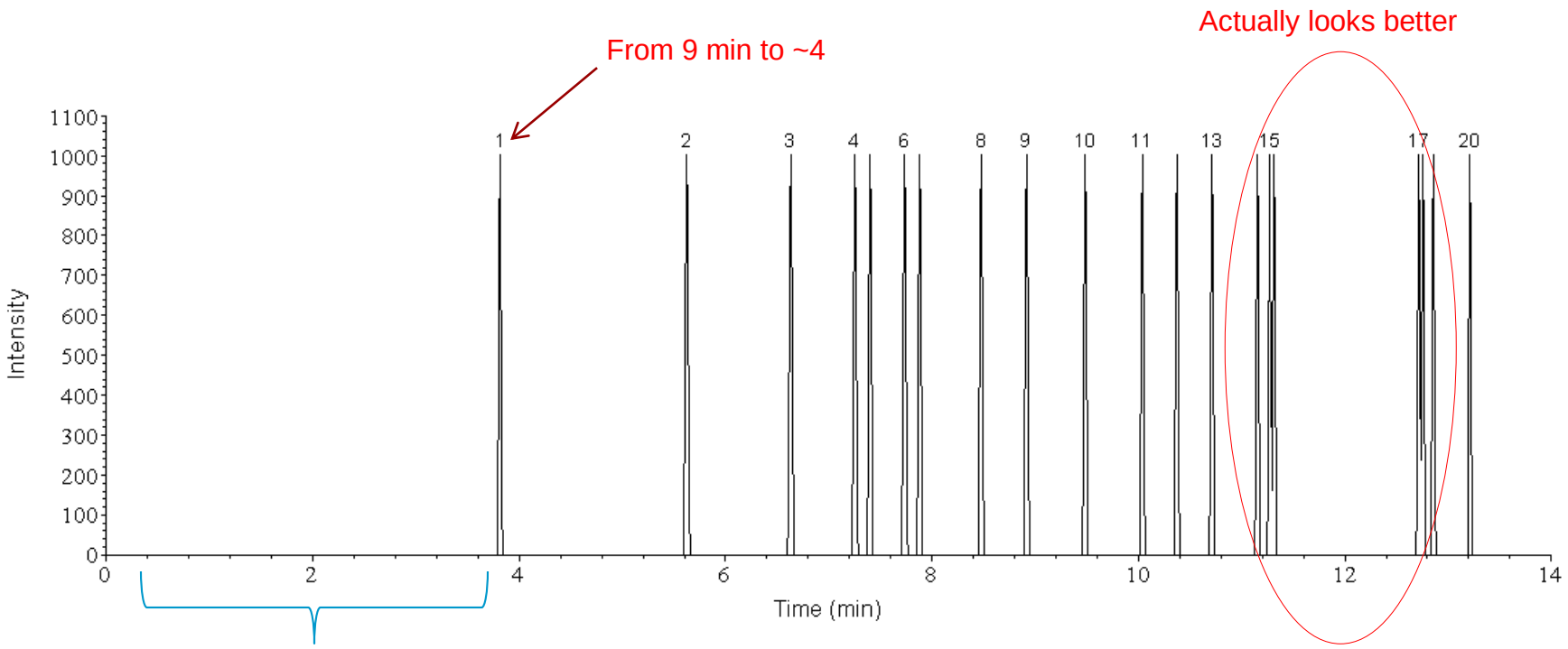
50°C for 5 minutes

10°C to 325°C

1st Peak comes out at 90°C

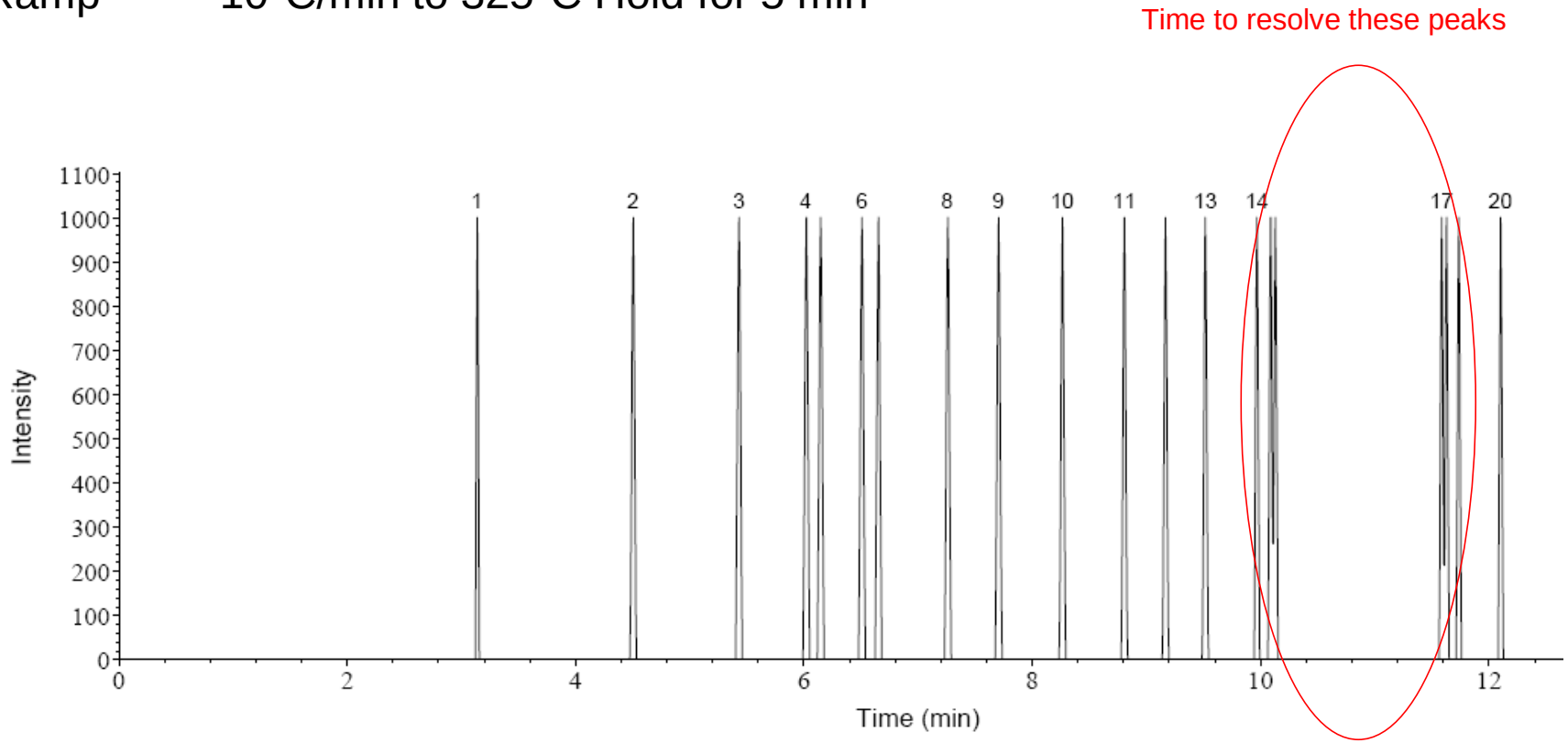
Developing Temperature Program
2nd Try

Initial Temp 90°C Hold for 5 min
Ramp 10°C/min to 325°C Hold for 5 min



Developing Temperature Program
3rd Try

Initial Temp 100°C Hold for 5 min
Ramp 10°C/min to 325°C Hold for 5 min



Resolve Co-elutions

Add a hold 20-30^{o} below the elution temperature

Co-elutions occur at 10 minutes

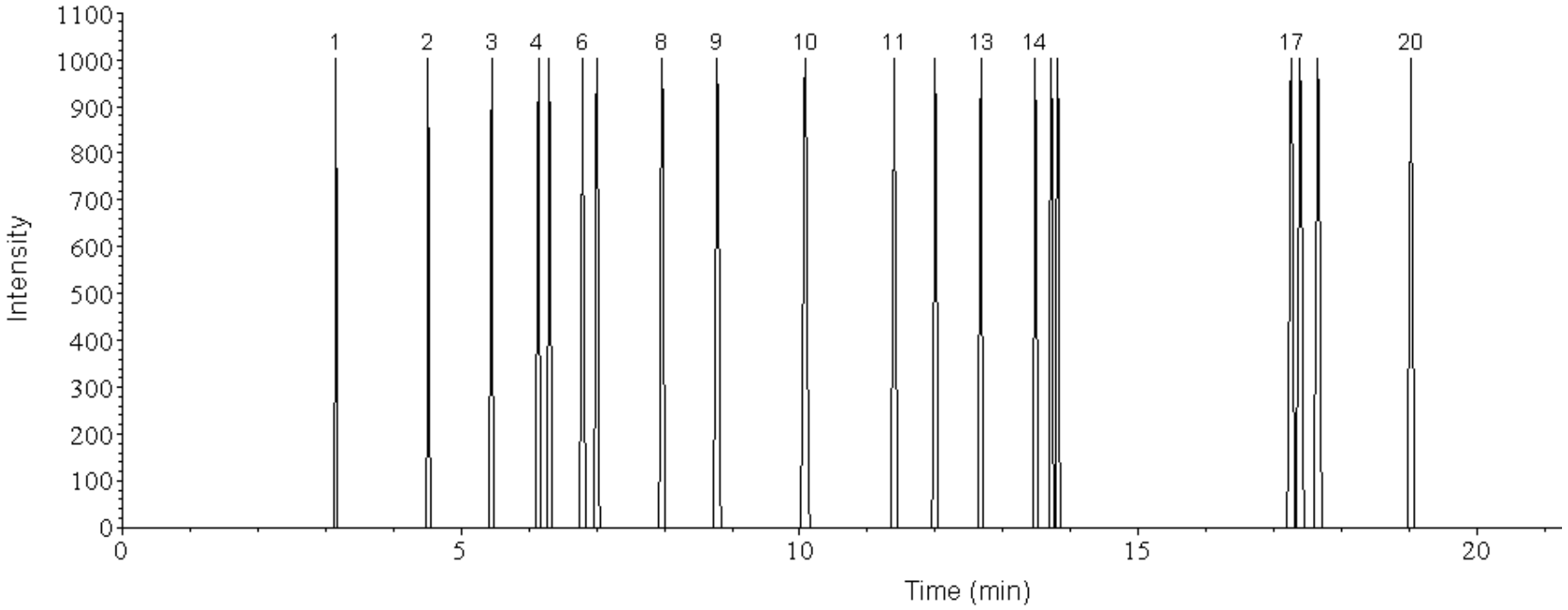
100^{o}C hold for 5 minutes
10^{o}C/min to 325^{o}C

Co-elutions occur at 150^{o}C

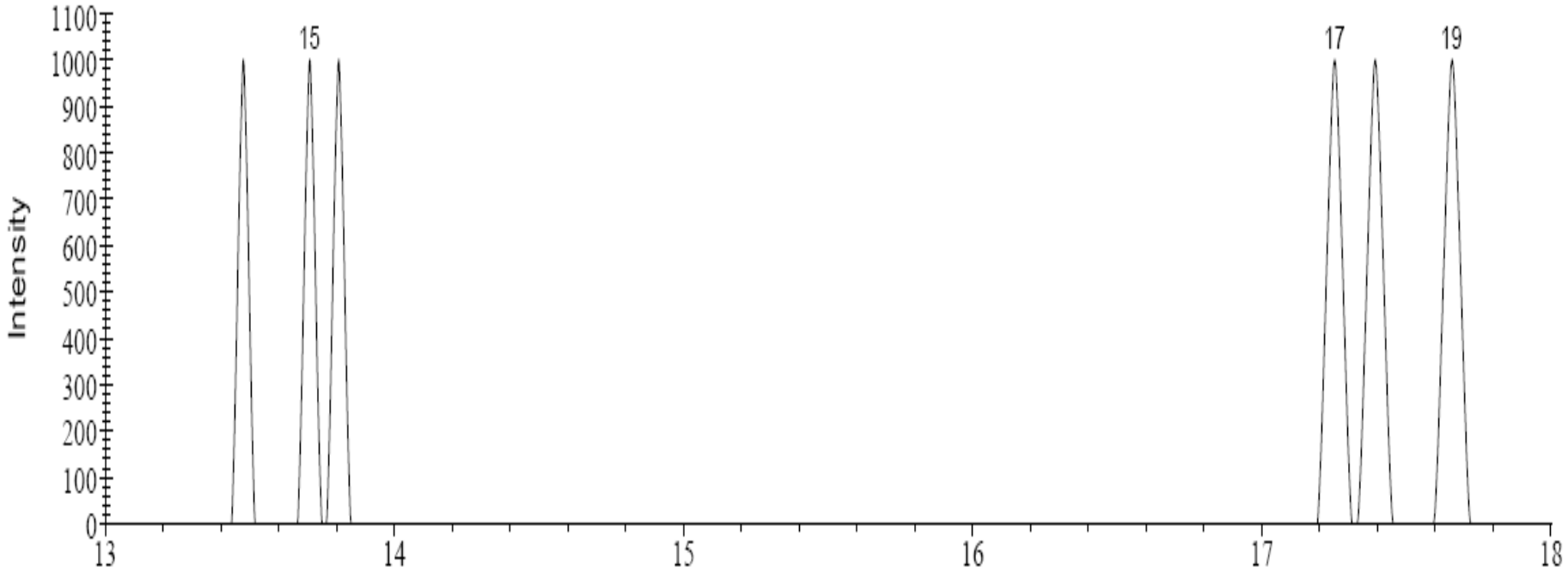
Set hold at 130^{o}C

Developing a Temperature Program

Oven: 100°C Hold for 5 minutes
10°C/min to 130°C hold for 5 min
10°C/min to 325°C



Developing a Temperature Program



Conclusions:

Think about the sample first

- **Is it chromatographable by GC?
- sample composition
- sample clean up
- level of detection

Use information sources first when choosing a column

Mild oven program to begin with

Utilize Technical Support

Conclusions: Starting Parameters

--Assuming S/Sl – FID system

Inlet Temp: 250°C

Split 50:1

Carrier Gas: Helium ~ 30 cm/sec, Hydrogen ~45 cm/sec

Oven Temp: 40°C hold for 5 minutes

10°C/min Ramp to Isothermal Limit of column

hold for 5-10 minutes

Detector Temp: 20°C above the highest oven temp

Questions

Contact Agilent Chemistries and Supplies Technical Support

1-800-227-9770 Option 3, Option 3:

Option 1 for GC/GCMS Columns and Supplies

Option 2 for LC/LCMS Columns and Supplies

Option 3 for Sample Preparation, Filtration and QuEChERS

Option 4 for Spectroscopy Supplies

Available in the USA 8-5 all time zones

gc-column-support@Agilent.com

lc-column-support@agilent.com

spp-support@agilent.com

spectro-supplies-support@agilent.com

Additional Resources and Application Support

Sample preparation eSeminar Series

https://www.agilent.com/en-us/training-events/eseminars/sample-preparation

Reference Materials and Guides:

Agilent Enhanced Matrix Removal – Lipid Brochure (Publication Number: 5991-6052EN)

https://www.agilent.com/cs/library/brochures/EMR%20Brochure%20CPOD%20Final_LoResSglPgs.pdf

https://www.agilent.com/en-us/products/sample-preparation/sample-preparation-methods/sample-preparation-methods/enhanced-matrix-removal-lipid

Agilent Sample Preparation Landing Page

https://www.agilent.com/en-us/products/sample-preparation/sample-preparation-methods

Agilent Sample Preparation Catalog (Publication Number: 5991-1057EN)

http://www.agilent.com/cs/library/catalogs/public/5991-1057EN%20Sample%20Prep%20Catalog.pdf