SUPELCO

595 North Harrison Road Bellefonte, PA 16823-0048 USA Telephone 800-247-6628 • 814-359-3441 Fax 800-447-3044 • 814-359-3044 email: supelco@sial.com sigma-aldrich.com/supelco

Bulletin 933

Capillary GC Column Choices for Residual Solvent Analyses Using Direct Injection or Solid Phase Microextraction* (SPME)

There is a variety of solvents used in pharmaceutical processing. In the process of preparing a pharmaceutical product, you can potentially retain residual organic solvents in the final preparation. In the interest of safety for the patient, the trend has been to use less toxic solvents during the manufacture of pharmaceutical preparations. In this bulletin, we discuss the choices of capillary GC columns most suitable for residual solvent analyses. We present the results of using the traditional direct injection technique, as well as, the fast, solvent free, and economical technique of SPME.

In the United States, the regulations require that you examine most pharmaceuticals to confirm the absence or very limited presence of many solvents. Current guidelines published by the International Conference on Harmonization of Technical Requirements for Registration of Pharmaceuticals for Human Use (ICH) describes a list of specific solvents, along with daily exposure limits. (1) The guidelines classify these solvents based on their toxicity:

- Class I: Solvents to be avoided
- Class II: Solvents to be limited
- Class III: Solvents with low toxic potential
- Solvents for which no adequate toxicological data is available

The compound lists for Classes I, II, and III contain 61 different solvents. No single column is capable of separating them all. For this reason, both of the analytical methods outlined by the United States and European Pharmacopoeia (USP and EP) describe the use of several capillary columns of different chemistries (2,3). We compared three columns that are equivalent to those described in both the EP and USP methods for the analysis of 60 of the 61 solvents (those detectable by GC/FID). Table 1 outlines the descriptions of these columns.

We analyzed the solvents by direct injection as three separate mixtures, divided by their individual classes (I, II or III). We prepared these standards specifically for this application. They are available through our Custom Chemical Standards Program.



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We can tailor these standards in combinations and concentrations to meet your specific needs. The run conditions were the same for all three columns. As expected, the elution order of the solvents varied for each column. The different elution orders are due to differences in chemical and physical properties of the solvents (boiling points, polarizability, dipole moments, number of hydrogen donor and hydrogen acceptor sites) and the strengths of the stationary phase-analyte interactions as described in Tables 2 and 3. The type and strength of these interactions determines the amount of time you will retain the analyte on the column.

Table 1. USP and EP Column Designations and Supelco Equivalents

Method	Column Designation	Supelco Equivalent
USP <467>, Method I	G27	Equity-5, cat. # 28279-U and precolumn, cat. # 25339
USP <467>, Methods IV&V	G43	OVI-G43, cat. # 25396 and precolumn cat. # 25339
USP <467>, Method VI	Various	Includes SUPELCOWAX 10, cat.# 25301-U
EP Method 2.4.24 - Primary column	6% polycyanopropyl phenylsiloxane	OVI-G43, cat. # 25396
EP Method 2.4.24 – Secondary column	Macrogol 20000	SUPELCOWAX 10, cat.# 25301-U

Table 2. Stationary Phase–Analyte Interactions

Interaction Type	Effect on Selectivity
Dispersive	elution by boiling point
π-π	elution by number of π -bonds
Dipole-induced dipole	elution by polarizability elution by dipole moment
Dipole-dipole	elution by dipole moment
Hydrogen bonding	elution by number of H-bond donor and/or acceptor sites

Table 3. Stationary Phase–Residual Solvent Interactions

Column	Type of Interaction
Equity-5	dispersive dipole-induced dipole
OVI-G43	π−π dispersive dipole-induced dipole dipole-dipole
SUPELCOWAX 10	π–π dispersive H-bonding dipole-dipole

We show chromatograms of each class of solvents on the three columns (Figures A through C) and information on the identity, retention time, and concentration of each peak (Tables 4 through 6). The advantage of having multiple columns available, of different selectivity, becomes evident when examining the information in these tables. A coelution on the primary analytical column will often be resolved on a secondary or confirmation column. For example, ethyl ether and ethanol, which coeluted on the OVI-G43, were resolved on both the Equity-5 and SUPELCOWAX 10. Under the run conditions used, a dual column analysis on the Equity-5 and SUPELCOWAX 10 will resolve all 60 solvents. Pairs not resolved on the OVI-G43 will be resolved on either the SUPELCOWAX 10 or the Equity-5. Since the run conditions were kept the same for all three columns, this makes it possible to do a single analysis of a solvent mixture by running two columns at the same time in a single GC oven. We suggest having all three columns available in your laboratory. The most suitable primary column for a particular analysis can be selected by studying Tables 4 through 6 and Figures A through C. Likewise, a second column can be chosen that will provide valuable confirmation information. This will guarantee success in being able to analyze any combination of solvents from the ICH list.

Table 4. Retention Times and Elution Order of Class I Residual Solvents on the Equity-5, SUPELCOWAX 10 and OVI-G43 Columns

Peak #	Identification	Concentration (µg/mL)	Retention Time Equity-5	Retention Time SUPELCOWAX 10	Retention Time OVI-G43
1	1,1-Dichloroethylene	4000	7.35	2.97	7.21
2	1,1,1-Trichloroethane	5000	18.11	5.54	17.35
3	1,2-Dichloroethane	2500	18.29	17.55	19.22
4	Carbon tetrachloride	2000	19.50	5.54	18.12
5	Benzene	1000	19.50	7.78	19.06

Table 5. Retention Times and Elution Order of Class II Residual Solvents on the Equity-5, SUPELCOWAX 10 and OVI-G43 Columns

Peak #	Identification	Concentration (µg/mL)	Retention Time Equity-5	Retention Time SUPELCOWAX 10	Retention Time OVI-G43
1	Methanol	1000	3.54	6.04	4.56
2	Acetonitrile	1000	5.91	11.80	8.30
3	Methylene chloride	1000	8.06	7.60	8.84
4	Nitromethane	250	9.94	22.91	NI
5	Hexane	1000	12.98	2.24	11.08
6	cis-1,2-Dichloroethylene	1000	14.00	10.97	14.74
7	Chloroform	300	15.16	13.28	16.67
8	2-Methoxyethanol	250	16.54	23.59	19.38
9	1,2-Dimethoxyethane	500	18.79	7.06	19.38
10	Cyclohexane	1000	19.56	2.87	17.54
11	Ethylene glycol	1000	20.62	36.92	29.44
12	Formamide	1000	20.62	40.49	29.90
13	Trichloroethylene	400	22.27	10.97	21.73
14	1,4-Dioxane	1000	22.69	16.31	23.13
15	2-Ethoxyethanol	800	22.89	25.37	24.47
16	Methylcyclohexane	1000	23.92	3.30	22.34
17	Pyridine	1000	24.91	23.59	25.80
18	Toluene	1000	26.44	14.28	26.12
19	Dimethylformamide	1000	26.75	28.99	29.73
20	Methyl butyl ketone	250	27.33	18.01	28.32
21	Chlorobenzene	1000	30.48	25.15	30.35
22	Dimethylacetamide	1000	31.01	31.24	33.76*
23	m-Xylene	333	31.41	21.75	31.08
24	p-Xylene	333	31.41	21.41	31.08
25	o-Xylene	333	32.46	23.83	32.21
26	n-Methylpyrrolidone	5000	37.73	38.07	39.93
27	Tetralin	500	42.33	34.70	41.98
28	Sulfolane	800	43.34	49.53	46.53

Peak #	Identification	Concentration (µg/mL)	Retention Time Equity-5	Retention Time SUPELCOWAX 10	Retention Time OVI-G43
1	Ethanol	3000	4.90	7.56	6.29
2	Acetone	3000	6.04	4.06	6.58
3	2-Propanol	3000	6.23	7.26	7.92
4	Pentane	3000	6.23	2.06	5.99
5	Ethyl ether	3000	6.64	2.32	6.29
6	Ethyl formate	3000	7.16	7.06	7.39
7	Methyl acetate	3000	7.78	4.31	8.54
8	1-Propanol	3000	9.50	14.76	12.60
9	Methyl-t-butyl ether	3000	10.33	2.65	9.92
10	Acetic acid	3000	NI	33.02	21.53
11	2-Butanone	3000	12.57	6.14	15.01
12	sec-Butanol	3000	13.07	13.45	16.47
13	Ethyl acetate	3000	14.68	5.78	15.58
14	Tetrahydrofuran	3000	16.21	4.93	16.35
15	iso-Butanol	3000	16.21	19.23	19.39
16	n-Butanol	3000	19.48	22.19	22.02
17	Isopropyl acetate	3000	19.48	6.14	19.89
18	Heptane	3000	22.20	2.65	20.47
19	Propyl acetate	3000	23.11	9.71	23.52
20	Isoamyl alcohol	3000	24.33	24.91	26.16
21	4-Methyl-2-pentanone	3000	24.68	11.62	25.72
22	n-Amyl alcohol	3000	26.13	26.49	27.73
23	Isobutyl acetate	3000	26.54	12.34	26.84
24	Butyl acetate	3000	28.51	17.72	28.79
25	Dimethyl sulfoxide	3000	29.18	35.51	33.02
26	Anisole	3000	33.30	29.65	33.54
27	Cumene	3000	33.68	23.38	33.29

Table 6. Retention Times and Elution Order of Class III Residual Solvents on the Equity-5, SUPELCOWAX 10 and OVI-G43 Columns.

NI: not integrated

Figure A. Class I Solvents on the Equity-5, SUPELCOWAX 10 and OVI-G43

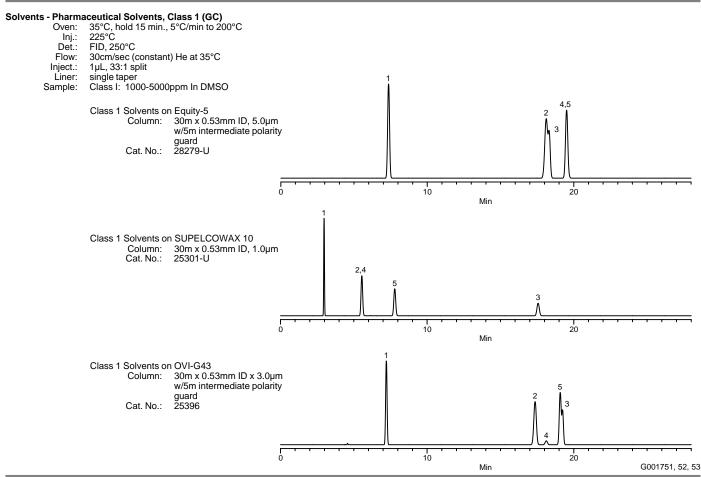


Figure B. Class II Solvents on the Equity-5, SUPELCOWAX 10 and OVI-G43

Solvents - Pharmaceutical Solvents, Class 2 (GC) Oven: 35°C, hold 15 min, 5°C/min to 200°C Inj.: 225°C

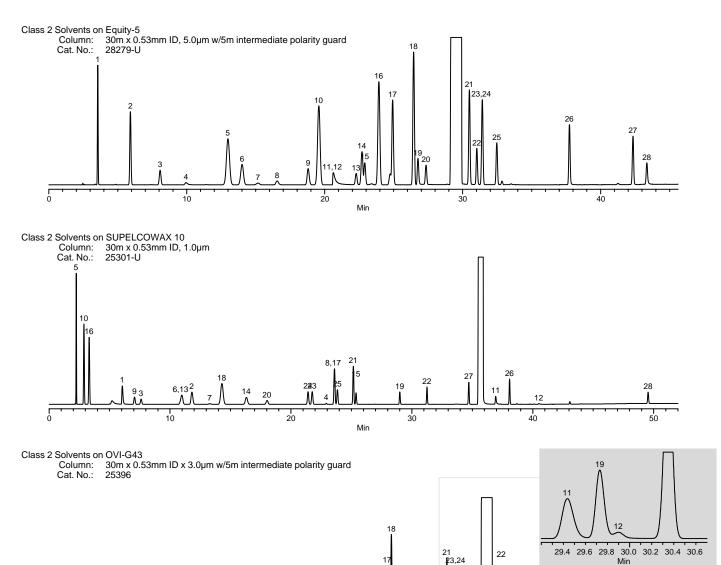
Inj.:

FID, 250°C 30cm/sec (constant) He at 35°C Det.: Flow:

Inject .: 1µL, 33:1 split

Liner: single taper

Class II and n-methylpyrrolidone, 250-1000ppm in DMSO Sample:



16

Min

10

3

10

8.9

20

26

40

25

30

27

28

G001754, G001755, G001756

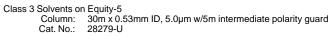
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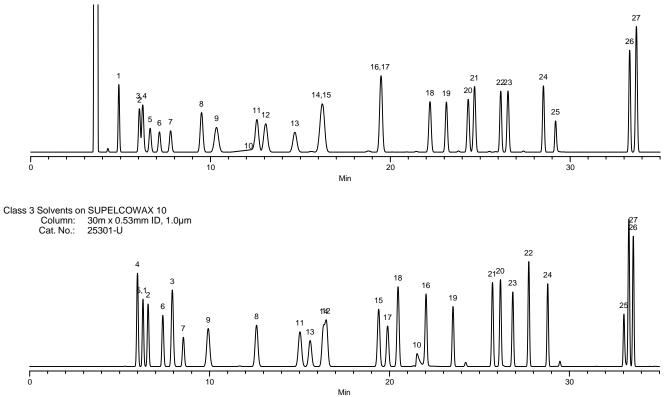
Figure C. Class III Solvents on the Equity-5, SUPELCOWAX 10 and OVI-G43

Solvents - Pharmaceutical Solvents, Class 3 (GC)

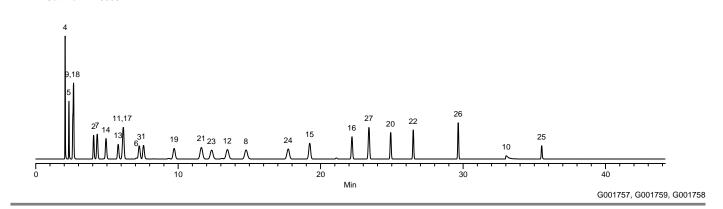
Oven:	35 C, hold 15 min, 5 C/min to 200 C
Inj.:	225°C
Det.:	FID, 250°C
Flow:	30cm/sec (constant) He at 35°C
Inject.:	1µL, 33:1 split
Liner:	single taper
Sample:	Class III: approx. 3000ppm in MeOH







Class 3 Solvents on OVI-G43 Column: 30m x 0.53mm ID x 3.0 μ m w/5m intermediate polarity guard Cat. No.: 25396

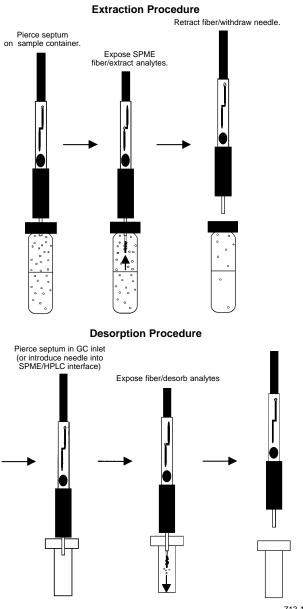


Fast GC Analysis of Residual Solvents using SPME with Dual Capillary GC Columns

The analysis of residual solvents can be time consuming and not always effective. As shown in the previous section of this bulletin, cycle time for the GC analysis of residual solvents is about 45 minutes. This analysis time is long if multiple samples need to be evaluated daily. Typically, analysts use either a static headspace method or direct injection of the sample for quantifying residual solvents. Both of these methods require long, high capacity columns. When such columns are used, analysis times can range from 30 to 60 minutes depending on the analytes you monitor. The method described below uses short narrow bore columns coupled with SPME. This technique provides the resolution you need with analysis times of less than 10 minutes.

SPME is an excellent alternative to headspace analyzers and direct injections. With SPME, you insert a coated fiber into the headspace of the vial and the analytes are concentrated onto the

Figure D. Solid Phase Microextraction



fiber. After a given amount of time (usually 5 min) the fiber is retracted into a needle, removed from the vial and inserted into the GC injection port where the fiber is desorbed (See Figure D). The fiber immediately releases the analytes into the injection port and onto the analytical column.

A low volume liner increases linear velocity and delivers the analytes onto the column with little or no band broadening. Therefore, you can use narrow bore columns. Since narrow bore columns provide more plates per meter than larger bore columns, you can use shorter length columns. This greatly reduces the analysis time while providing resolution similar to longer columns.

As previously mentioned in this bulletin, no single column is capable of separating a complex mixture of analytes classified as residual solvents; therefore, a second column is required. In this procedure, we installed a nonpolar Equity-1 column and an intermediate polarity VOCOL column in one injection port. These columns provide distinct differences in polarity while having compatible temperature ranges. We matched the column flows by shortening the column with the longer methane retention time. When the methane retention times between the two columns were within 10% of each other, the columns were ready for separating analytes.

For the evaluation of SPME and dual column separations, we prepared the solvents by class and extracted them from water using SPME. We determined that water was the best solvent choice for extraction of the solvents. However, if the drug or finished product is not water-soluble, you can dissolve it in a solvent such as DMSO and spike it into a water sample containing 25% sodium chloride. When extracting polar solvents, it is best to add salt to the sample. You can add the salt after you dissolve the sample and you can adjust the pH to enhance extraction efficiency.

Table 7 shows the best conditions for extracting the analytes in each class. We were able to extract most analytes at concentrations of 5ppm or less using these conditions.

Table 7. Residual Solvent Extraction andAnalysis Conditions

Extraction Con	ditions for Class 1			
Sample: Extraction: Desorption:	100µm PDMS 1ppm each analyte in 2mL water with 25% NaCl heated headspace, 50°C for 5 min in 4mL vial 3 min at 250°C			
Extraction Co	nditions for Class II			
Sample: Extraction: Desorption:	85µm Polyacrylate 5ppm each analyte in 2mL water & 25% NaCl, pH 11 heated headspace, 60°C for 5 min in 4mL vial 3 min at 250°C			
Extraction Co	nditions for Class III			
Sample: Extraction:	100µm PDMS 5ppm each analyte in 2mL water & 25% NaCl, pH 2 heated headspace, 60°C for 5 min in 4mL vial 3 min at 250°C			
	-			
	Equity-1 and VOCOL both 10m x 0.20mm ID x 1.2µm 40°C (hold 0.75 min) to 200°C at 20°C/min (hold 10 min)			
	helium, 35cm/sec @40°C (9psi constant pressure) Split 5:1@40°C, 0.75mm liner, 2 columns in 1 port using 0.8mm graphite ferrule			
Detector:				

We extracted the analytes using heated headspace from an aqueous matrix containing 25% salt. We recommend two types of fiber coatings. The 100 μ m PDMS fiber is suitable for Class I and Class III solvents. Many analytes in the Class II list are polar,

which requires a more polar fiber for extraction, therefore, we selected the $85\mu m$ polyacrylate fiber. Figures E through J show the chromatograms of the analytes extracted by SPME.

Figure E. Class I Solvents on Equity-1 Column Using SPME (1ppm each in water)



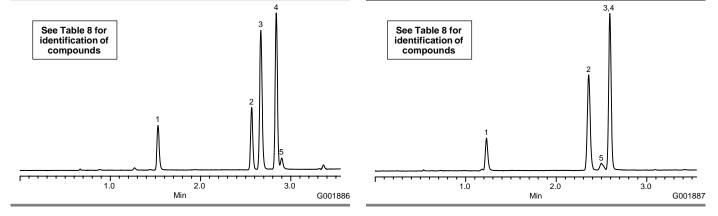


Figure G. Class II Solvents on Equity-1 Column Using SPME (5ppm each in water)

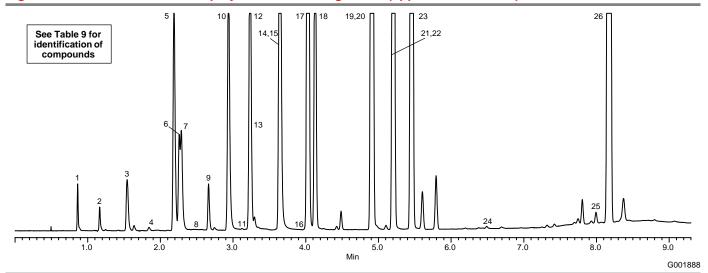
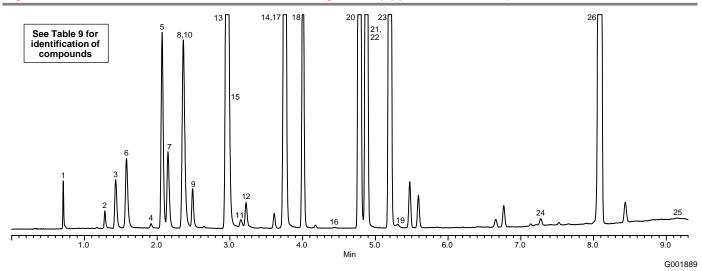
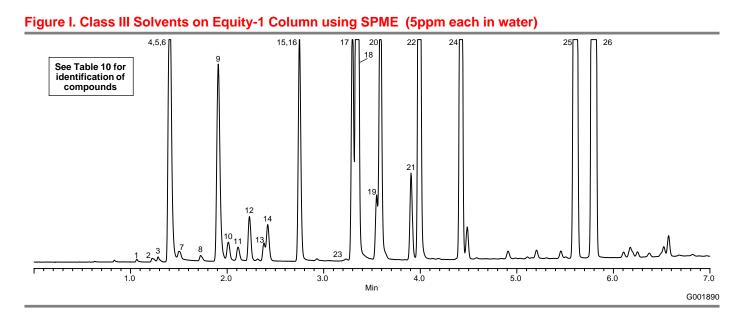
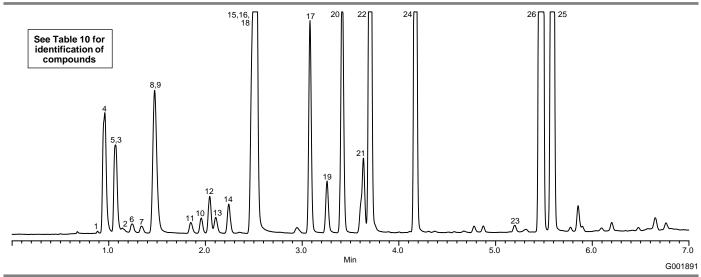


Figure H. Class II Solvents on VOCOL Column Using SPME (5ppm each in water)









SPME is excellent for extracting nonpolar and moderately polar analytes. Only polar analytes with low vapor pressures are difficult to extract. Some of the peaks look very small, but that is in comparison to some nonpolar analytes that you can easily extract. By reducing the intensity scale, one can see that the peaks are sufficiently large and symmetrical for proper quantification.

Some of the residual solvents such as ethylene glycol and formamide have very low vapor pressures and you cannot analyze these by headspace. Therefore, we have not shown these in the chromatograms. Other aprotic and polar solvents such as dimethylacetamide, DMSO, sulfolane, n,n-dimethylformamide and glycol ethers are difficult to analyze by headspace. You can use SPME to extract these analytes, but the minimum quantitation limits will be higher than for less polar analytes. You usually analyze these analytes by direct injection due to their low vapor pressures. By immersing the SPME fiber directly into the aqueous matrix, you can achieve slightly lower detection limits relative to headspace SPME.

Tables 8 through 10 show a listing of all of the analytes in the various classes along with the concentration that we used during the SPME evaluation. In the tables, we make a recommendation to the applicability of SPME for detection and quantification of the analytes. R is for recommended by SPME, D is for difficult by SPME, and N is for not recommended by SPME. We base these recommendations on a minimum quantitation limit of 5ppm or less. In cases where D is listed, it would be difficult to quantify the analyte at the concentration listed, but at higher concentration levels, these compounds should be quantifiable using SPME.

Table 8. Class I Solvents Using SPME

Peak #	Identification Class I	Concentration (µg/mL)	SPME Use	Retention Time Equity-1	Retention Time VOCOL
1	1,1-Dichloroethene	1	R	1.54	1.32
2	1,1,1-Trichloroethane	1	R	2.58	2.36
3	1,2-Dichloroethane	1	R	2.66	2.63
4	Benzene	1	R	2.82	2.63
5	Carbon tetrachloride	1	R	2.89	2.51

Table 9. Class II Solvents Using SPME

Peak #	Identification Class II	Concentration (µg/mL)	SPME Use	Retention Time Equity-1	Retention Time VOCOL
1	Methanol	5	R	0.87	0.69
2	Acetonitrile	5	R	1.17	1.26
3	Methylene chloride	5	R	1.52	1.56
4	Nitromethane	5	R	1.64	1.89
5	Hexane	5	R	2.26	2.04
6	cis-1,2-Dichloroethylene	5	R	2.18	2.12
7	Chloroform	5	R	2.34	2.35
8	2-Methoxyethanol	5	D	2.43	2.30
9	1,2-Dimethoxyethane	5	R	2.66	2.54
10	Cyclohexane	5	R	2.94	2.36
11	2-Ethoxyethanol	5	D	3.30	3.12
12	1,4-Dioxane	5	R	2.97	3.21
13	Trichloroethene	5	R	3.23	2.85
14	Pyridine	5	R	3.62	3.77
15	Methylcyclohexane	5	R	3.66	2.98
16	Dimethylformamide	5	D	3.87	4.39
17	Toluene	5	R	4.03	3.77
18	Methyl butyl ketone	5	R	4.14	4.02
19	Dimethylacetamide	5	D	4.88	5.60
20	Chlorobenzene	5	R	4.92	4.81
21	p-Xylene	5	R	5.21	4.20
22	m-Xylene	5	R	5.21	4.20
23	o-Xylene	5	R	5.46	5.24
24	n-Methylpyrrolidone	5	D	6.66	7.25
25	Sulfolane	5	D	8.14	9.12
26	Tetralin	5	R	8.18	8.14
-	Ethylene glycol	1000	N		-
	Formamide	1000	N		
=recommende	ed; D=difficult; N=not recommende	d			

Table 10. Class III Solvents using SPME

Peak #	Identification Class III	Concentration (µg/mL)	SPME Use	Retention Time Equity-1	Retention Time VOCOL
1	Ethanol	5	R	1.10	0.92
2	Acetone	5	R	1.25	1.15
3	2-Propanol	5	R	1.32	1.07
4	Pentane	5	R	1.39	0.96
5	Ethyl ether	5	R	1.41	1.07
6	Methyl acetate	5	R	1.52	1.34
7	Ethyl formate	5	R	1.43	1.24
8	1-Propanol	5	R	1.75	1.52
9	Methyl-t-butyl ether	5	R	1.92	1.47
10	2-Butanone	5	R	2.03	1.96
11	sec-Butanol	5	R	2.13	1.84
	Acetic acid	5	D	2.18	1.73
12	Ethyl acetate	5	R	2.25	2.04
13	Tetrahydrofuran	5	R	2.39	2.24
14	iso-Butanol	5	R	2.43	2.11
15	n-Butanol	5	R	2.75	2.49
16	Isopropyl acetate	5	R	2.76	2.50
17	Propyl acetate	5	R	3.31	3.09
18	Heptane	5	R	3.35	2.52
19	Isoamyl alcohol	5	R	3.56	3.28
20	4-Methyl-2-pentanone	5	R	3.60	3.43
21	n-Amyl alcohol	5	R	3.91	3.64
22	Isobutyl acetate	5	R	3.99	3.69
23	Butyl acetate	5	R	4.43	4.19
24	Dimethyl sulfoxide	5	D	4.31	5.10
25	Anisole	5	R	5.61	5.63
26	Cumene	5	R	5.80	5.51
=recommende	d; D=difficult; N=not recommended				

We determined that 57 of 60 analytes can be analyzed by SPME, with eight of these being somewhat difficult to extract at 5ppm. SPME is a good alternative to conventional headspace analysis because of the short analysis time and good recovery for the majority of the residual solvent analytes.

Using SPME for Quantitative Analysis of Residual Solvents

Many pharmaceutical companies use SPME on a routine basis for residual solvent analyses. They have demonstrated reproducible and quantitative results using SPME. The work of Scypinski and Smith at Hoffmann-La Roche Inc. (Nutley, New Jersey, USA) demonstrated the use of SPME for quantitative analysis of residual solvents. Their work compared headspace SPME and immersion SPME for determining residual solvents in several water-soluble drug substances (4).

Immersion and headspace SPME were essentially equal with respect to precision, sensitivity, and accuracy (Table 11). The Hoffmann-La Roche chemists preferred the headspace method because it prolonged the lifetime of the SPME fiber. A 100µm polydimethylsiloxane-coated fiber provided higher sensitivity toward the nonpolar analytes (i.e., the residual solvents). A polyacrylate-coated fiber offered higher sensitivity toward the polar analytes (alcohols). Using the polydimethylsiloxane-coated fiber, detection limits ranged from 0.06µg/mL and 0.3µg/mL for 1,4-dioxane (by headspace and immersion, respectively) to 0.002µg/mL for benzene (both techniques). For their analysis, they added methanol at 1.0% v/v in the water diluent to obtain reproducible residual solvent results. Based on these results, the chemists concluded that the SPME sample introduction technique is useful for screening residual solvents in pharmaceutical drug substances.

Because liquid and headspace sampling methods differ in kinetics, you should consider the two approaches complementary. For a given sampling time, other analysts have found immersion SPME is more sensitive than headspace SPME for analytes predominantly present in the liquid (5). The reverse was true for analytes that reside primarily in the headspace. These generalizations can be used to your advantage to selectively adsorb more volatile or less volatile compounds, as a situation warrants. For higher sensitivity from headspace SPME, the sample headspace should be as small as is practical. Zhang and Pawliszyn present a detailed theoretical discussion of headspace SPME in reference 6.

SPME is fast, easy, economical, and eliminates the costs and hazards associated with using organic solvents. Under consistent sampling conditions, you can extract analytes with good precision over wide ranges of concentrations. Good precision also makes the technique effective in quantitative analyses. If you are interested in reducing the time and expense of sample concentration in your analyses, SPME is the ideal answer to your needs.

Table 11. Precision and Detection Limits of SPME/Capillary GC for Organic Volatile Impurities and Final Recrystallization Solvent

	Precision (% RSD)		Detection L	imit (µg/mL)
	Headspace	Immersion	Headspace	Immersion
Acetone	1.1	0.5	0.2	0.4
Ethanol	7.0	5.8	5.0	2.0
Isopropanol	1.4	1.9	0.6	1.6
Benzene	2.7	2.8	0.002	0.002
Chloroform	3.2	2.2	0.03	0.04
1,4-Dioxane	1.9	2.2	0.06	0.3
Methylene chloride	2.6	2.2	0.06	0.08
Trichloroethene	3.4	3.2	0.02	0.01
Data from reference 4.				

References

1. ICH Guidance for Industry, Q3C Impurities: Residual Solvents, US Department of Health and Human Services Food and Drug Administration, Center for Drug Evaluation and Research (CDER), Center for Biologics Evaluation and Research (CBER), ICH December 1997.

2. USP 25, NF20, <467>, pgs. 1943-1944.

- 3. European Pharmacopoeia 4, 2.4.24, pgs. 96-100.
- 4. Rosen Shaw, S., A. M. Smith, L. Clark Nelson, and S. Scypinski, poster presentation, American Association of Pharmaceutical Science Conference, June 1994.
- 5. Yang, X. and T. Peppard, J. Agric. Food Chem. 42: 1925-1930 (1994).
- 6. Zhang, Z. and J. Pawliszyn, Anal. Chem. 65: 1843-1852 (1993).

References not available from Supelco.

Patents

*Solid Phase Microextraction (SPME) Technology licensed exclusively to Supelco. US patent #5,691,206; European patent #523092.

Trademarks

Equity, SUPELCOWAX, VOCOL, Supelco - Sigma-Aldrich.

Ordering Information

Equity-1	Length (m)	d _r (μm)	Beta	Cat. No.
0.10mm ID	15	0.10	250	28039-U
0.20mm ID	12	0.33	152	28041-U
	25	0.33	152	28042-U
	10	1.2	42	28043-U
0.25mm ID	30	0.10	625	28044-U
	15 30	0.25 0.25	250 250	28045-U 28046-U
	60	0.25	250	28047-U
	15	1.0	63	28048-U
	30 60	1.0 1.0	63 63	28049-U 28050-U
	100	1.0	63	28050-0 28052-U
0.32mm ID	30	0.10	800	28053-U
	15	0.25	320	28054-U
	30	0.25	320	28055-U
	60 30	0.25 1.0	320 80	28056-U 28057-U
	60	1.0	80	28058-U
	100	1.0	80	28060-U
	30 30	2.0 5.0	40 16	28061-U 28062-U
	60	5.0	16	28063-U
0.53mm ID	15	0.10	1325	28064-U
	30	0.10	1325	28065-U
	15 30	0.5 0.5	265 265	28067-U 28068-U
	15	1.0	133	28069-U
	30	1.0	133	28071-U
	15 30	1.5 1.5	88 88	28072-U 28073-U
	60	1.5	88	28073-0 28074-U
	15	3.0	44	28075-U
	30 60	3.0 3.0	44 44	28076-U 28077-U
	15	5.0	27	28079-U
	30	5.0	27	28081-U
	60	5.0	27	28082-U
Equity-5	Length (m)	d _r (μm)	Beta	Cat. No.
0.10mm ID	15	0.10	250	28083-U
0.20mm ID	15	0.20	250	28084-U
0.201111110	30	0.20	250	28085-U
	60	0.20	250	28086-U
	12	0.33	152	28087-U
0.25mm ID	15	0.25	250	28088-U
	30 60	0.25 0.25	250 250	28089-U 28090-U
	30	0.5	125	28092-U
	15	1.0	63	28093-U
	30 60	1.0 1.0	63 63	28094-U 28095-U
0.32mm ID	15	0.25	320	28096-U
01021111112	30	0.25	320	28097-U
	60 30	0.25	320	28098-U
		0.32	250	28099-U 28195-U
	30 30	0.5 1.0	160 80	28199-U
	30 30 60	0.5 1.0 1.0		
0.53mm ID	30 30 60 15	0.5 1.0 1.0 0.5	80 80 265	28199-U 28251-U 28252-U
0.53mm ID	30 30 60 15 30	0.5 1.0 1.0 0.5 0.5	80 80 265 265	28199-U 28251-U 28252-U 28259-U
0.53mm ID	30 30 60 15	0.5 1.0 1.0 0.5	80 80 265	28199-U 28251-U 28252-U
0.53mm ID	30 30 60 15 30 60 30 15	0.5 1.0 1.0 0.5 0.5 0.5 1.0 1.5	80 80 265 265 265 133 88	28199-U 28251-U 28252-U 28259-U 28263-U 28263-U 28264-U 28265-U
0.53mm ID	30 30 60 15 30 60 30 15 30	0.5 1.0 1.0 0.5 0.5 0.5 1.0 1.5 1.5	80 80 265 265 265 133 88 88	28199-U 28251-U 28252-U 28269-U 28263-U 28264-U 28265-U 28267-U
0.53mm ID	30 30 60 15 30 60 30 15	0.5 1.0 1.0 0.5 0.5 0.5 1.0 1.5	80 80 265 265 265 133 88	28199-U 28251-U 28252-U 28259-U 28263-U 28263-U 28264-U 28265-U
0.53mm ID	30 30 60 15 30 60 30 15 30 30 60 15	0.5 1.0 0.5 0.5 1.0 1.5 1.5 3.0 3.0 5.0	80 80 265 265 133 88 88 44 44 44 27	28199-U 28251-U 28252-U 28263-U 28263-U 28265-U 28265-U 28267-U 28268-U 28268-U 28269-U 28278-U
0.53mm ID	30 30 60 15 30 60 30 15 30 30 30 60	0.5 1.0 1.0 0.5 0.5 1.0 1.5 3.0 3.0	80 80 265 265 133 88 88 88 44 44	28199-U 28251-U 28259-U 28263-U 28263-U 28265-U 28265-U 28267-U 28268-U 28269-U

SUPELCOWAX 10						
Length (m)	d _r (µm)	Beta	Cat. No.			
0.10mm ID Fused Silica						
5	0.10	250	25025-U			
10	0.10	250	25026-U			
15	0.10	250	24343			
0.20mm ID Fused Silica	0.20	250	24169			
30 60	0.20	250 250	24169			
0.25mm ID Fused Silica	0.20	200	24110			
0.25mm ID Fused Silica 15	0.25	250	24077			
30	0.25	250	24079			
60	0.25	250	24081			
30	0.50	125	24284			
0.32mm ID Fused Silica						
15	0.25	320	24078			
30	0.25	320	24080-U			
60 15	0.25 0.50	320 160	24082 24083			
30	0.50	160	24084			
60	0.50	160	24085-U			
30	1.0	80	24211			
60	1.0	80	24212			
0.53mm ID Fused Silica						
15	0.50	265	25324			
30 60	0.50 0.50	265 265	25325 25385			
15	1.0	133	25300-U			
30	1.0	133	25301-U			
60	1.0	133	25391			
30	2.0	63	25375-U			
60	2.0	63	25376			
Solvents: OVI-G43						
Length (m)	d _, (μm)	Beta	Cat. No.			
0.53mm ID Fused Silica						
30	3.0	44	25396			
Deactivated Guard Column for OVI-G43						
5m x 0.53mm ID			25339			

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