# **Determination of Zn in River Water**

### **■** Introduction

In Notification No. 123 from Japanese Ministry of Environment (November 5, 2003), part of the environmental standards related to water pollution was amended. The amendment establishes total zinc as an additional environmental standard for maintenance of the environment, in particular, for aquatic life and its habitat and maintenance of a growth environment in public water regions.

The standard is applicable to specified water regions. The criteria value differs depending on the type of water body, with 0.03mg/L or less for rivers and lakes, and 0.02 or 0.01mg/L or less for coastal waters.

For the applicable analytical method, the industrial wastewater test method (JIS K0102) is used, including the following methods.

- 1. Flame atomic absorption method
- 2. Furnace atomic absorption method
- 3. ICP emission spectrometric method
- 4. ICP mass spectrometric method

The present study examines the determination of zinc in river water standard samples JAC 0031 (non-spiked) and JAC 0032 (spiked), provided by the Japan Society for Analytical Chemistry, using flame atomic absorption (hereafter, flame method) and furnace atomic absorption (hereafter, furnace method).

## ■ Instrument and Analytical Conditions

Tables 1 and 2 show the instrument and analytical conditions used with each method, respectively. In the furnace method, analysis sensitivities at low concentration and high concentration was changed by changing the flow rate of the argon gas passing through the graphite tube during atomization.

Table 1 Analytical Conditions in Flame AA

| AA-6300                    |
|----------------------------|
| 213.8nm                    |
| 0.7nm                      |
| 10mA                       |
| BGC-D2                     |
| Air – acetylene            |
| 7mm                        |
| Upper limit 0.04mg/L (ppm) |
|                            |

#### Table 2 Analytical Conditions in Furnace AA

| rable 2 Analytical Conditions in Furnace AA |  |  |  |  |  |
|---|--|--|--|--|--|
| Instrument                                  | Main unit AA-6800,                           |  |  |  |  |
|   | Atomization unit GFA-EX7                     |  |  |  |  |
| Analysis wavelength                         | 213.8nm                                      |  |  |  |  |
| Slit width                                  | 0.5nm  |  |  |  |  |
| Current                                     | 10mA   |  |  |  |  |
| Lamp mode                                   | BGC-D2                                       |  |  |  |  |
| Tube type                                   | High density graphite tube                   |  |  |  |  |
| Smpl. Inj. Vol.                             | 2 – 20μL (total injection volume: 20μL)      |  |  |  |  |
|   | Drying 120°C                                 |  |  |  |  |
| Taman Dramman                               | Ashing 400°C                                 |  |  |  |  |
| Temp. Program                               | Atomization 2000°C                           |  |  |  |  |
|   | Cleaning 2500°C                              |  |  |  |  |
| Chal Cal Cama                               | Low concentration: Upper limit 1.5µg/L (ppb) |  |  |  |  |
| Std. Sol. Conc.                             | High concentration: Upper limit 15μg/L (ppb) |  |  |  |  |

## ■ Results

The methods used for adjusting the sensitivity in atomic absorption analysis include changing the analysis wavelength, and in the flame method, adjusting the burner angle. However, in the furnace method, sensitivity adjustment is possible by adjusting the argon gas flow rate during atomization. Fig.1 shows the relation between the gas flow rate and the absorbance of zinc at 2ppb.

Fig.2 displays the calibration curve of the flame method. Fig.3 shows the low concentration range calibration curve of the furnace method in which the argon gas flow rate was 0.0L/min. Fig.4 shows the high concentration range calibration curve of the furnace method in which the argon gas flow rate was 0.5L/min.

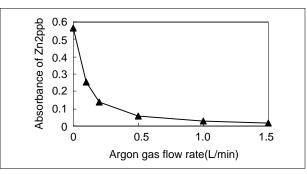


Fig.1 Relation between Ar Flow Rate and Sensitivity

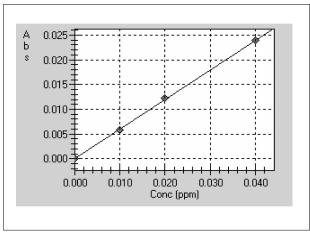


Fig.2 Calibration Curve in Flame Method

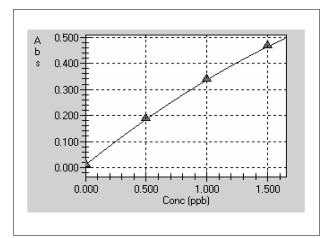


Fig.3 Calibration Curve in Furnace Method at Low Concentrations

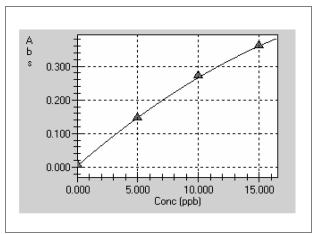


Fig.4 Calibration Curve in Furnace Method at High Concentrations

Fig.5 shows the peak profiles of the standard solutions at low concentrations in the furnace method.

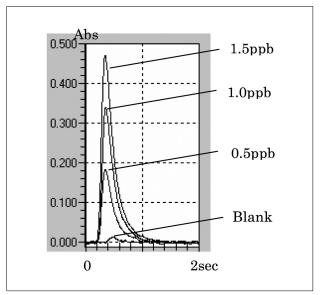


Fig.5 Peak Profiles in Furnace Method

Table 3 presents the results of analysis. Table 4 shows the expected lower quantification limits in the test solution in the flame method and furnace method, respectively, based on the results of the analysis. The lower quantification limit in the furnace method was predicted based on the analysis results in the low concentration range.

**Table 3 Analysis Results** 

|                          | Certified value   | Flame AA |         | Furnace AA<br>(High Conc.) |
|--------------------------|-------------------|----------|---------|----------------------------|
| JAC 0031<br>(non-spiked) | 0.00079<br>(mg/L) | <0.007   | 0.00079 | 0.00074                    |
| JAC 0032<br>(spiked)     | 0.0113<br>(mg/L)  | 0.011    | 0.0105  | 0.0106                     |

JAC 0032 was analyzed at low concentrations using the furnace method, with the injection volume decreased to  $2\mu L.$ 

## **Table 4 Quantification Limits**

| Flame Method   | 0.007mg/L  |
|----------------|------------|
| Furnace Method | 0.0001mg/L |

In the flame method, since direct quantification of zinc below 0.01mg/L is difficult, concentration processing is necessary. On the other hand, in the furnace method, quantification is possible down to 0.0001mg/L, 1/100 of the standard value, without boosting the concentration. However, with samples like seawater that have high salt concentrations, it is necessary to be careful of spectral and chemical interferences. In addition, with zinc concentrations lower than 0.01mg/L, it is necessary to take adequate precautions against the contamination from the containers or the acid used during processing.

