Approaching The Ultimate Limits of Detection For Endocrine Disruptors in Wastewater Effluent Using GC-NCI-MS/MS

Melissa Churley¹, Ruth-Marfil-Vega², Marc Mills³ and Anthony Macherone^{4,5} ¹Agilent Technologies, Inc., Santa Clara, CA;² American Water, Belleville, IL;³ National Risk Management Research Laboratory (NRMRL), US EPA, Cincinnati, OH;⁴Agilent Technologies, Inc., Wilmington, DE;⁴ ⁵The Johns Hopkins School of Medicine, Department of Biological Chemistry, Baltimore, MD, USA

Objective

To efficiently evaluate our wastewaters, an integrated approach combining bioassays with targeted chemical analysis needs to be implemented. High-throughput, robust and sensitive analytical methods are required.

Described herein is the development of a rugged GC-NCI-MS/MS analytical method designed for detecting trace levels of Endocrine Disrupting Chemicals (EDCs) in decreased sample volumes (< 500mL) of wastewater effluent

EDC analysis by GC-NCI-MS/MS appears to be a plausible alternative to LC-MS/MS analysis based in part on the potential advantage to use smaller sample volumes, thus lowering testing costs, and achieving similar or lower detection levels. In fact, estimated MDLs for estrogens and androgens are at least two orders of magnitude lower than those determined by LC/MS/MS.

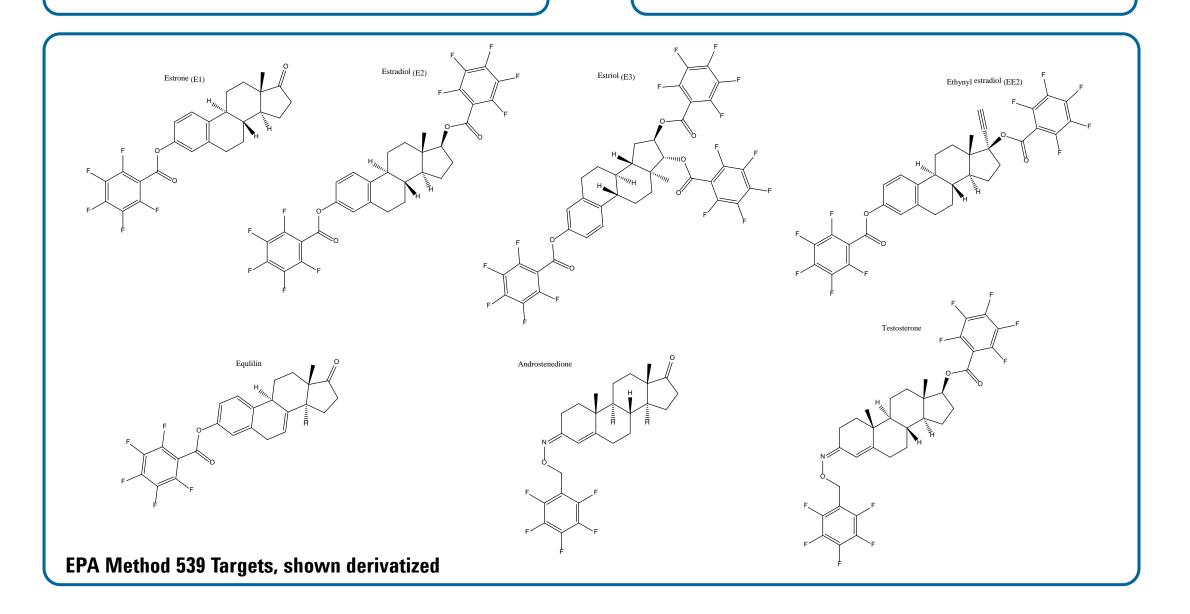
This work centers on EPA Method 539, currently an LC-ESI-MS/MS method (with derivatization required), in an effort to evaluate the GC-NCI/MS/MS approach.

Experimental

Analysis was performed using an Agilent 7890A Gas Chromatograph coupled to a 7000B Triple Quadrupole Mass Spectrometer equipped with a Multi Mode Inlet and Purged Ultimate Union used for back flushing the column. The union was placed between two DB-17ht columns of dimensions 5m x 0.25mm x 0.15µm and 15m x 0.25mm x 0.15 μ m. The GC was programmed to reach 330°C in constant flow mode and 2 µL was injected. Back flush was accomplished by lowering the inlet head pressure for 1 minute post run.

Negative chemical ionization was performed with 40% ammonia reagent gas and a source temperature of 150°C.The MS source temperature was 310°C. An EM gain of 100 was used for this analysis. Derivatization was carried out to produce either the pentafluorobenzoyl (PFB) ester or PFB oxime at C-3. The PFB ester was formed at C-17 in some cases and also at C-16 in the case of E3. Calibration was performed using non-extracted standards with levels ranging from 0.05 pg/mL to 10 ng/ml. ISTDs Estradiol-D3, Testosterone-D5 and Estrone-D4 were added at 10 pg/ml.

Extracted 20 ml and 500 ml wastewater samples were obtained from the US EPA.



Experimental

Derivatization Procedure

- Add 0.5 mL 1% pyridine in ethyl acetate to dried sample 8.
- Add 50 µL 10% pentafluorobenzoyl chloride in ethyl acetate
- Cap and vortex 1 minute, heat 30 min at 60°C
- Evaporate to dryness at 60°C under gentle N_2 stream
- Add 1 mL 0.5M aqueous sodium bicarbonate, vortex
- 6. Cap and react 10 minutes at room temperature
- Centrifuge 5 min

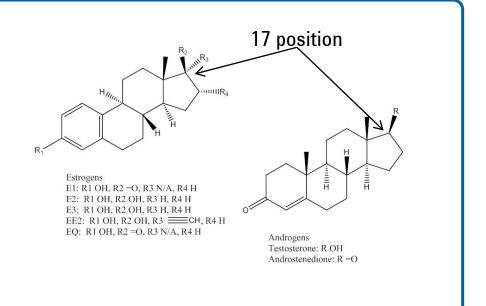
- Transfer organic layer to a second tube for next steps
- Dry at 60°C under gentle nitrogen a. If estrogens only, go to step 14
- 10. Add 0.5 mL isooctane
- 11. Add 100 μ L 0.1% (wt/v) PFB-hydroxylamine hvdrochloride in pyridine
- 12. Cap and vortex 1 minute, heat at 60°C for 5 min
- 13. Dry at 60°C under gentle nitrogen
- 14. Add 50 uL isooctane for injection

Notes about derivatization chemistry

All estrogens react with pentafluorobenzoyl chloride at R positions where R = OH to form the corresponding ester.

In all cases (estrogen and androgen) where the 17 position is a keto moiety, pentafluorobenzyl hydroxylamine either reacts poorly or not at

Using this derivatization procedure, all estrogens have a predictable SRM transition $M \rightarrow (M-64)$.



Time Segment	Compound	Precursor Ion	Product Ion	dwell	Collision Energy
1	EE2	490.5	426.5	60	6
1	E1	464.4	400.4	60	6
1	EQ	462.4	398.4	60	4
1	EQ	462.4	370.4	60	8
1	ASD	461.5	431.5	60	6
2	Testosterone	677.6	657.6	100	4
2	Testosterone	677.6	627.6	100	6
2	E2	660.5	596.5	100	8
3	E3	870.6	806.6	150	6
3	E3	870.6	167.6	150	10

Table of SRM transitions





Results and Discussion

Results

The previously reported limit of detection for E2 was again achieved in our laboratory (Churley and Macherone, 2010.; Macherone, 2012). Figure 1 illustrates the S/N ration for 8 replicate injections of E2.

Figure 1. Eight replicate injections of E2.

8 Noise (RMS)= 42.41; SNR (11.442min)=525.0	22202.5	
Noise (RMS)=20.19; SNR (11.448min)=1099.4	22 180 6	
Noise (RMS)= 15.24; SNR (11.442min)= 1384.3	21)00.5	
Noise (RMS)= 14.43; SNR (11.448min)= 1428.0	20902 6	
Noise (RMS)= 18.73; SNR (11.452min)= 1202.0	22500.5	
Noise (RMS)= 18.31; SNR (11.453min)= 1255.5	22907.7	
Noise (RMS)= 19.58; SNR (11.463min)= 1094.0	21468.8	
Noise (RMS)= 15.05; SNR (11.467min)= 1437.3	21940	

The instrument detection limit (IDL) can be determined by the equation:

 $IDL = (t_a)(\%RSD)(amount of standard)/100$

where t_a is a statistical confidence factor found in the Student t- distribution table. With 99% confidence, the t value for 8-1 degrees of freedom is 2.998. Substituting this into equation 1 gives:

IDL = (2.998)(3.0%)(10 pg/ml)/100 = 0.9 pg/ml E2

Table 1 illustrates the raw data for the IDL determination of estradiol (E2) and Table 2 illustrates the area precision for a 2 μ l injection of a 0.5 μ g/ml E2 standard.

Table 1. Raw E2 data for IDL determination			Table 2. Area precision for		
Es	tradiol	E2 at 0.5 pg/ml (2 μl injection = 1 fg on			
Inection #	S/N	Area	column)		
1	483	20005	Name	E2 Area	
2	509	20515	0p5 01.D	684	
3	405	20722	0p5 02.D	620	
4	548	21139	0p5_02.D	723	
5	420	21368	0p5_03.D	641	
6	352	21525	0p5_04.D	620	
7	399	21614	Average	658	
8	410	21894	St Dev	45	
Average	441	21098	% RSD	6.8	
St Dev		636	70 HSD	0.0	
% RSD		3.0			

Conclusions

The accurate and sensitive measurement of steroidal analogs is an important requirement to monitor the fate and transport of steroidal analogues in the environment. This poster outlines a procedure that modifies steroidal analogs such that they become amenable to electron capture negative chemical ionization mechanisms and provides GC/MS/MS conditions required for a highly sensitive and robust analytical method with IDL on the order of 1 pg/ml or less. Moreover, the procedures described herein are amenable to matrices other than waste water effluent, soil and bio-solids and can be extended to biological sources such as serum or plasma for measurement of the exposome. In addition to E2 data, preliminary data for 17a-ethinylestradiol (EE2) in wastewater show that calibration is possible using a set of standards containing from 0.2 to 200 pg per vial, where $R^2 = 0.99$. Based on a 20 mL sample volume, this would be equivalent to 0.01 to 10 ng/L. An extracted 20 mL wastewater sample was quantified to contain EE2 at 17 ng/L (ppt) with adequate GC peak separation.

References

A. Macherone (2012),

Agilent Application Note 5990-9478EN M. Churley and A. Macherone (2010).

Agilent Application Note 5990-5513EN

M. Churley, A. Macherone and R. White, (2011) 59th ASMS Conference on Mass Spectrometry and Allied Topics.

Acknowledgements

The authors would like to gratefully acknowledge the United States Environmental Protection Agency and especially Ruth Marfil-Vega and Marc Mills for providing standards and prepared samples for the study presented herein.

The United States Environmental Protection Agency through its Office of Research and Development contributed to the research described here. Mention of trade names or commercial products does not constitute endorsement or recommendation for use by USEPA.

