Using the JetClean Self-Cleaning Ion Source to Extend Maintenance Free Operation

Bruce Quimby¹, Michael Szelewski², Elizabeth Almasi¹, Ken Brady¹ ¹Agilent Technologies Inc., Wilmington, DE, USA ²M Szelewski LLC, Hockessin, DE, USA



Introduction

Phthalate Analysis

Matrix and column bleed deposits degrade instrument response over time, necessitating routine source cleaning. The process requires removal and abrasive cleaning of the source, reducing productivity. The patented JetClean self-cleaning ion source (JetClean) was developed to maintain consistent MS response in difficult matrices for extended periods of time via controlled introduction of hydrogen into the MS source. This poster describes the application of JetClean to the analysis of phthalate esters (phthalates). Due to the phthalates' adverse health effects (supspected endocrine disruptors) their use is limited by international regulations.

Phthalates can exhibit several undesirable characteristics in GC/MS analysis:

- Non-linearity: some analytes have reduced response at the lower cal levels.
- <u>Peak Tailing</u>: compounds stick to the source with some ions tailing more than others.

Results

Peak Shape Improvement

The first parameters investigated were source temperature and drawout lens diameter. A diameter of 9 mm instead of the typical 3 mm and a temperature of 300°C were found to be optimum. However, without hydrogen cleaning gas added, the peaks still showed tailing. The degree of tailing was not the same on all ions, EICs for multiple ions from the same peak had differing amounts of tailing.

The figure to the right shows that this

9 mm Drawout lens Source 300°C $0.3 \text{ mL/min H}_2 \text{ added}$ **Green Trace FID**

- Sensitivity: Higher source temperatures are often used to improve linearity, but some compounds lose sensitivity.
- **Dropping Response:** Raw area response for replicate injections can exhibit significant loss of response with time.
- These problems were investigated and addressed with a number of hardware and method changes, resulting in significantly improved results.

Experimental

			Instru	ment Condi	tions and Test Mi	xture			
GC	Agilent	7890E	3				MSD		Agilent 5977 Xtr
Ramp Initial Ramp 1 Ramp 2 Runtime	°C/min °C Hold min 60 1.0 20 220 1.0 5 290 4.0 40 min		Inlet Liner Column	Ultra Inert Single Taper w/ Wool (PN. 5190-2293) HP-5MSUI (19091S-433UI) 30m x 0.25 mm id x 0.25 um		Solvent Acquisit Tune EMV Quad Te	Delay ion Mode mp	5.0 min SIM, TID ON Atune Gain 1 150 'C	
nlet Split/Sp Temp 290 °C Node Hot Pul nit. Pressure 15.0 psi Pulse Pressure 35 psi u		olit/Splitless (or MMI) 0 °C ot Pulsed Splitless, .0 psi psi until 1.0 min		Column Flow Splitter Pressure MSD Restictor	1.0 mL/min, const flow 2-way splitter 4.2 psig (initial) 0.68 m x 0.12 mm id 1.2 mL/min const flow		Source 1 Transfer Drawout JetClean JetClean	Femp Line Lens Flow Mode	300 'C 280 'C 9 mm 0.3 mL/min H ₂ continuous
Purge Flow Septum Purge njection volume	50 mL/min at 1.0 min 3 mL/min, switched 0.5 uL			FID Restictor Split ratio	0.37 m x 0.12 mm id ~1:1 MSD:FID				
Hardware			I	Liquid	AUX EPC 1 Helium	FID		JetClean Hydrogen	

differential tailing is greatly reduced with the continuous addition of 0.3 mL/min of H₂ to the source, resulting in TIC and FID peak shapes to be very comparable.



Linearity Improvement

The calibration		Cal Range, pg	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5 - 1250
parameters were		Stat	r²	%RSD	r ²	%RSD	r ²	%RSD	r ²	%RSD
studied, examining	ļ	Source Temp °C	300	300	230	230	230	230	230	230
source temperature	e, H ₂	H ₂ , mL/min	0.3	0.3	0.3	0.3	0.3	0.3	no H ₂	no H ₂
addition, and tune		Tune	Atune	Atune	Atune	Atune	Etune	Etune	Etune	Etune
type. In all cases, th	he 9	Tetradecane	0.9999	3.0	0.9999	5.9	0.9999	4.2	0.9979	22.9
mm drawout lens w	was	Dimethyl	0.9998	2.4	0.9999	6.2	0.9993	6.9	0.9993	15.0
used. Calibrations		Diethyl	0.9998	1.9	0.9999	6.9	0.9990	8.8	0.9987	17.5
were ISTD with 3 ru	uns	Disobutyl	0.9998	4.4	0.9999	8.8	0.9987	9.4	0.9990	15.9
at each level.		Dibutyl	0.9999	10.9	0.9998	3 14.7	0.9983	14.0	0.9982	21.2
		Bis(2-methoxyethy	0.9998	21.1	0.9990) 21.9	0.9980	26.3	0.9980	35.8
Optimal results we	re	Bis(4-methylpentyl	0.9996	4.2	0.9999	9.5	0.9990	12.5	0.9991	20.0
found with the sour	irce	Bis(2-ethoxyethyl)	0.9997	16.7	0.9994	20.8	0.9982	24.6	0.9983	32.0
at 300°C, 0.3 mL/min H_2		Dipentyl	0.9998	3.8	0.9998	3 7.8	0.9978	15.7	0.9969	27.9
added, and using		Dihexyl	0.9998	4.1	0.9999	6.8	0.9984	13.3	0.9988	20.0
ATUNE.		Benzyl butyl	0.9997	5.9	0.9997	13.0	0.9982	19.7	0.9975	32.6
	< 0.999	Bis(2-butoxyethyl)	0.9998	13.8	0.9997	20.3	0.9982	23.0	0.9990	30.3
		Dicyclohexyl	0.9979	5.4	0.9989	23.6	0.9976	30.7	0.9971	42.3
	< 30%	Bis(2-ethylhexyl)	0.9997	3.8	0.9999	7.9	0.9981	13.7	0.9982	22.7
	< 20%	Di n-octyl	0.9997	2.9	0.9999	7.7	0.9984	14.1	0.9992	19.3
	× 10%	Dinonyl	0.9996	3.1	0.9998	9.8	0.9985	14.4	0.9993	20.0

Area Precision

The system has a CFT splitter between the MSD and FID. The FID signal makes a good reference for peak shape, linearity, and precision. Use of the FID helps distinguish between detector effects and inlet, liner, and column effects.

Configuration





One significant problem with phthalates is dropping response with replicate injections. This problem may be present even if no matrix is injected. Employing JetClean significantly increased precision, which was further improved by employing a Merlin Microseal instead of the standard spectrum.



Normalized area response of 60 consecutive injections of the phthalate mixture at 125pg level. All compounds show a remarkable replicate area precision. The Merlin Microseal contribution to the precision improvement is currently under investigation.

Summary and Conclusions

The GC/MS analysis of phthalates can be improved in terms of peakshape, linearity, and repeatability by incorporating these changes to the analysis method:

- Run the source temperature at 300°C. Values lower than that may result in tailing and dropping response. Higher values result in problems with some of the more thermally labile phthalates.
- Reduce drawout lens interactions with the phthalates by changing to a larger diameter, like 9 mm.
- Using ATUNE.U instead of ETUNE.U improves linearity and peakshape for phthalates.
- Addition of continuous hydrogen to the source during analysis with JetClean improves peakshape, linearity, and replicate precision.
- Use of a Merlin Microseal further improved response stability.

Cal Range, pg	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5 - 1250	2.5	- 1250	2.5 - 1250		2.5 - 1250	
Stat	r ²	%RSD	r ²	%RSD		r ²	%RSD		r ²	
Source Temp C	300	300	230	230 230		230	230		300	
H2, mL/min	0.3	0.3	0.3	0.3		0.3	0.3		100	
Tune	Atune	Atune	Atune	Atune	E	tune	Etune		Atune	
Tetradecane	0.9999	3.0	0.9999	5.9		0.9999	4.2		0.9999	
Dimethyl	0.9998	2.4	0.9999	6.2		0.9993	6.9		0.9999	
Diethyl	0.9998	1.9	0.9999	6.9		0.9990	8.8		0.9999	
Disobutyl	0.9998	4.4	0.9999	8.8		0.9987	9.4		0.9998	
Dibutyl	0.9999	10.9	0.9998	14.7		0.9983	14.0		0.9997	
Bis(2-methoxyethy	0.9998	21.1	0.9990	21.9		0.9980	26.3		0.9994	
Bis(4-methylpentyl	0.9996	4.2	0.9999	9.5		0.9990	12.5		0.9998	
Bis(2-ethoxyethyl)	0.9997	16.7	0.9994	20.8		0.9982	24.6		0.9996	
Dipentyl	0.9998	3.8	0.9998	7.8		0.9978	15.7		0.9996	
Dihexyl	0.9998	4.1	0.9999	6.8		0.9984	13.3		0.9997	
Benzyl butyl	0.9997	5.9	0.9997	13.0		0.9982	19.7		0.9995	
Bis(2-butoxyethyl)	0.9998	13.8	0.9997	20.3		0.9982	23.0		0.9995	
Dicyclohexyl	0.9979	5.4	0.9989	23.6		0.9976	30.7		0.9986	
Bis(2-ethylhexyl)	0.9997	3.8	0.9999	7.9		0.9981	13.7		0.9993	
Di n-octyl	0 9997	29	0 9999	77		0 9984	14 1		0 9995	

	- L Y I	0.5557	2.5	0.5555	/./	0.550+	14.1	0.5555
Dinon	y l	0.9996	3.1	0.9998	9.8	0.9985	14.4	0.9993