

# Considerations when working with solid samples and SPE

Solid phase extraction is routinely selected as a sample preparation technique prior to HPLC and GC analysis. It provides:

- Clean-up;
- Analyte pre-concentration;
- Solvent switching.

This works well for any homogeneous liquid samples, but clearly a solid sample cannot be applied directly to an SPE device. Here we will consider how best to approach the handling of solid samples prior to SPE.

The first stage of sample pre-treatment involves homogenising the solid sample with a suitable solvent. The solvent selected for this purpose will:

- Be effective at solubilising the analyte.
- Be effective at solubilising the sample matrix, although this is not necessary providing the sample is blended well with the solvent.
- Ideally it will be a weak solvent when it comes to the SPE stage. This may not be possible, and it may be necessary to dry down and reconstitute the sample prior to loading on to the SPE cartridge.

Having formed a suspension of the sample in a suitable solvent the sample should be vortexed to promote thorough mixing, and then centrifuged to pellet the insoluble components of the matrix. The solvent should then be removed for further treatment.

At this point the SPE mechanism needs to be selected:

## Reversed Phase SPE Method

Mechanism	Weak Solvent	Strong Solvent
Reversed phase	Water/buffer	Acetonitrile/Methanol (hexane strongest)
Normal phase	Hexane/Heptane	Mid polar organic such as DCM, MTBE, IPA (strongest is water)
Ion exchange	Aqueous buffer at a pH which promotes ion exchange interactions between analyte and SPE sorbent	Organic solvent, typically methanol with pH adjustment to disrupt the ion exchange interactions.

Analyte properties and sample diluent both need to be considered at this point. Reversed phase extractions are effective for moderately polar to moderately hydrophobic analytes, but only if they are dissolved in a weak solvent such as water. If the sample is dissolved in a mid-polar solvent the sample will either need to be diluted, or alternatively dried down and reconstituted. The same is true for normal phase, where samples need to be dissolved in hexane/heptane for analytes to be successfully retained during sample loading onto the SPE device.

These examples will hopefully illustrate this concept.

### **1) The extraction of PCBs from soil.**

PCBs are hydrophobic and will dissolve readily in a non-polar solvent. Soil can be dried, weighed and then extracted using hexane (or heptane). Once thoroughly mixed and centrifuged the resulting solution can be applied directed to a conditioned silica SPE tube allowing for a normal phase extraction to be conducted.

### **2) Tylosin from meat.**

Tylosin is an antibiotic, with a log P of 1.0. Extracting from meat with a mixture of acetonitrile and water (80:20) will break any hydrophobic interactions releasing the tylosin. After centrifugation the resulting liquid sample will require further dilution with water to reduce the loading solvent strength prior to a reversed phase extraction.

### **3) Enilconazole from oranges.**

Enilconazole is a fungicide widely used in agriculture and is often used with citrus fruits. It can readily be extracted from oranges using acetonitrile. As enilconazole is basic, the addition of HCl to the acetonitrile extract will promote protonation of the analyte, allowing for it to be directly retained on a strong cation exchange SPE material. This greatly simplifies the extraction process as no dilution of the sample is required (in contrast to reversed phase, or normal phase extraction where 100% acetonitrile would cause the analyte to be unretained during loading).

For further advice or suggestions regarding sample preparation please visit:

[www.phenomenex.com/techniques/solid-phase-extraction](http://www.phenomenex.com/techniques/solid-phase-extraction)