

APPLY THESE FUNDAMENTALS TO PERFORM RELIABLE AIR QUALITY TESTS

Together, the Agilent 5977/7890B Series GC/MSD and Markes TD systems allow you to take consistent air samples and confidently test them for a variety of contaminants.

MARKES
international



MEASURING VOLATILES IN AIR: US EPA METHOD TO-17

US Clean Air Act regulations have identified specific Hazardous Air Pollutants (HAPs), also known as air toxics. These analytes cover a wide range of polarities and volatilities, and are most effectively monitored using pumped sampling onto multisorbent tubes, followed by automated TD-GC/MS (scan) analysis.

TO-17-type methods, which are based on pumped air monitoring, facilitate the simultaneous analysis of non-polar and polar organic vapors – including volatile and semi-volatile components.



OZONE PRECURSORS IN AMBIENT AIR

C_2 to C_{10} hydrocarbons from car exhausts have been identified as precursors to the formation of street-level ozone and urban smog. US, European, and other regulators require round-the-clock monitoring of these compounds in major urban areas, particularly during the summer months.

In addition, regulations developed in response to the Kyoto protocol on greenhouse gases require the monitoring of trace-level ultra-volatile compounds with high global warming and ozone-depleting potential. These include perfluorinated hydrocarbons (such as CF_4 and C_2F_6), the tracer gas SF_6 , and N_2O . Unfortunately, these compounds begin to boil at $-128\text{ }^\circ\text{C}$ – and are extremely difficult to trap, concentrate, and measure at low levels; therefore, an online sampling system is required.

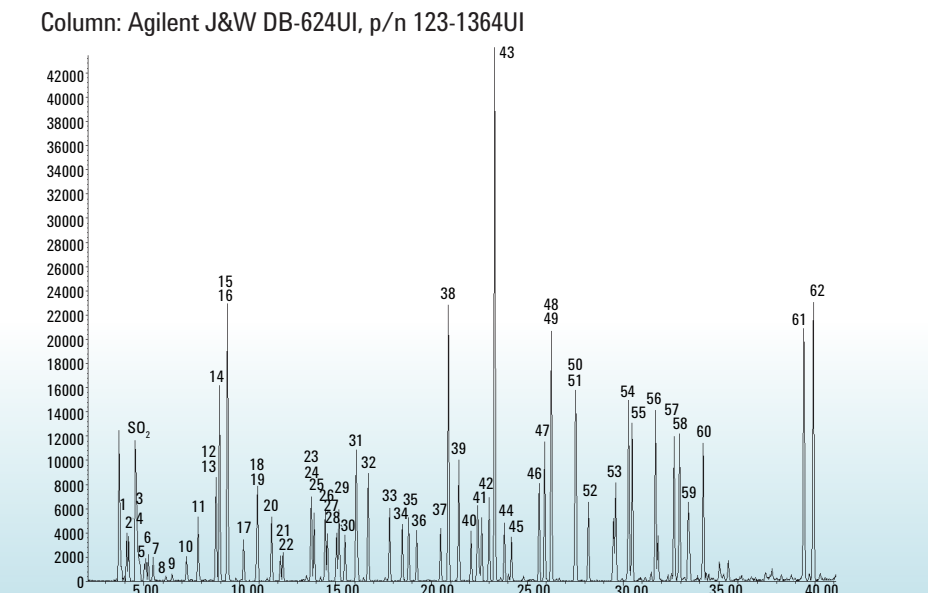


URBAN AIR TOXICS: TO-14 AND TO-15

EPA Methods TO-14 and TO-15 cover the testing of ambient air for toxic organic compounds. Generally, TO-14 is limited to the analysis of non-polar compounds, while TO-15 is larger in scope and better defined for analyzing VOCs in air and other gaseous matrices.

Canisters are ideally suited for sampling ultra-volatile organics, such as freons and C_2 hydrocarbons, which are difficult to trap on sorbent tubes at ambient temperature. They also make grab sampling more convenient.

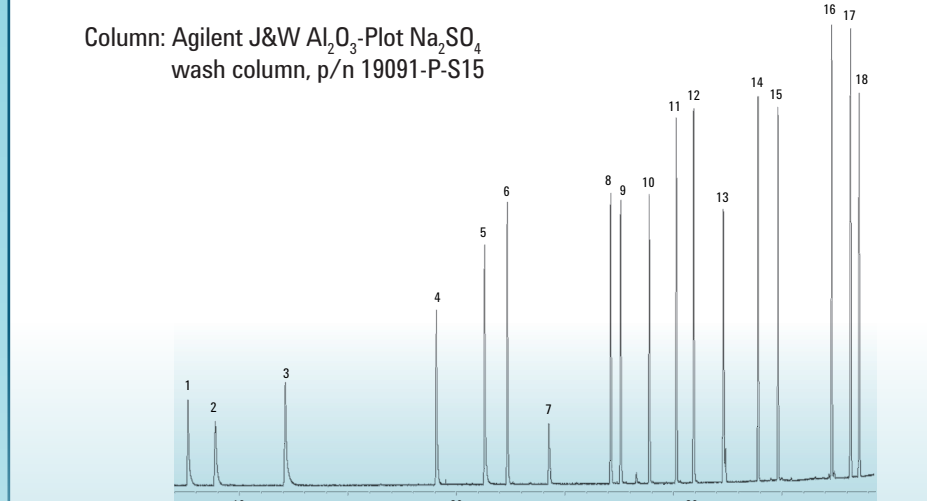
Air toxics in urban air



Column: Agilent J&W DB-624UI, p/n 123-1364UI
Analysis of urban air toxics: Splitless analysis of a 1-liter pumped sampling of air toxics standard (1 ppb) using ATA tubes. Note the extracted mass ion 45 for IPA, which demonstrates excellent peak shape. See Markes Technical note TDTS 86a for method details.

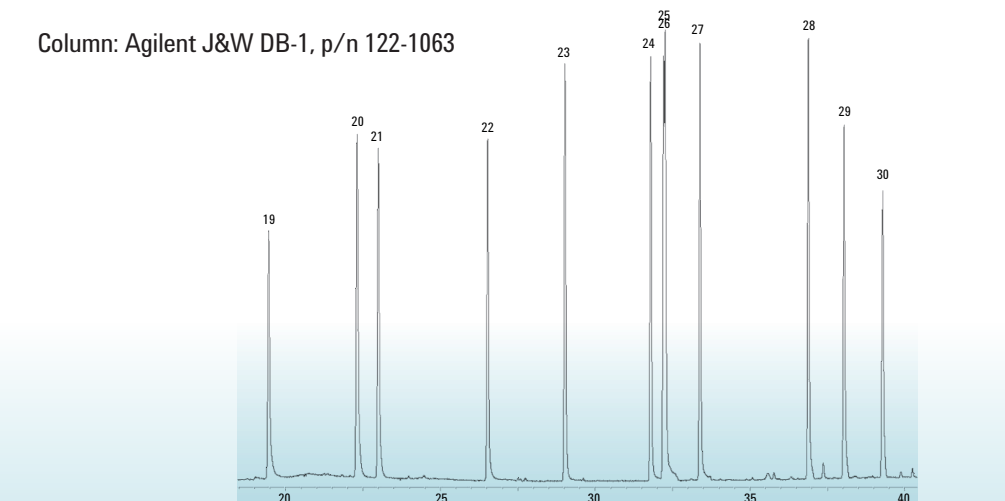
- | | | |
|------------------------------------|--------------------------------|-------------------------------|
| 1. Propylene | 22. Cis-1,2-Dichloroethylene | 44. Dibromochloromethane |
| 2. Dichlorodifluoromethane | 23. Methyl ethyl ketone | 45. 1,2-Dibromoethane |
| 3. 1,2-Dichlorotetrafluoroethane | 24. Ethyl acetate | 46. Chlorobenzene |
| 4. Methyl chloride | 25. Tetrahydrofuran | 47. Ethylbenzene |
| 5. 1,2-Dichloroethane | 26. Chloroform | 48. m-Xylene |
| 6. 1,3-Butadiene | 27. 1,1,1-Trichloroethane | 49. p-Xylene |
| 7. Vinyl chloride | 28. Cyclohexane | 50. o-Xylene |
| 8. Methyl bromide (bromomethane) | 29. Carbon tetrachloride | 51. Styrene |
| 9. Chloroethane | 30. Benzene | 52. Tribromomethane |
| 10. Trichlorofluoromethane | 31. n-Heptane | 53. 1,1,2,2-Tetrachloroethane |
| 11. Ethanol | 32. Trichloroethylene | 54. 1,2,4-Trimethylbenzene |
| 12. 1,2-Dichloroethylene | 33. 1,2-Dichloropropane | 55. 1,3,5-Trimethylbenzene |
| 13. 1,1,2-Trichlorotrifluoroethane | 34. 1,4-Dioxane | 56. 1-Ethyl-4-methyl benzene |
| (Freon 113) | 35. Bromodichloromethane | 57. 1,2-Dichlorobenzene |
| 14. Acetone | 36. Trans-1,3-dichloropropene | 58. 1,3-Dichlorobenzene |
| 15. Carbon disulfide | 37. Methyl isobutyl ketone | 59. alpha-Chloromethylbenzene |
| 16. Isopropyl alcohol | 38. Toluene | 60. 1,4-Dichlorobenzene |
| 17. Methylene chloride | 39. Cis-1,3-Dichloropropene | 61. 1,2,4-Trichlorobenzene |
| 18. Tert-butyl methyl ether | 40. Trans-1,2-Dichloroethylene | 62. Hexachloro-1,3-butadiene |
| 19. n-Hexane | 41. 1,1,2-Trichloroethane | 42. Tetrachloroethylene |
| 20. 1,1-Dichloroethane | 43. Methyl n-butyl ketone | |

Analysis of C_2 to C_{10} hydrocarbons in ambient air



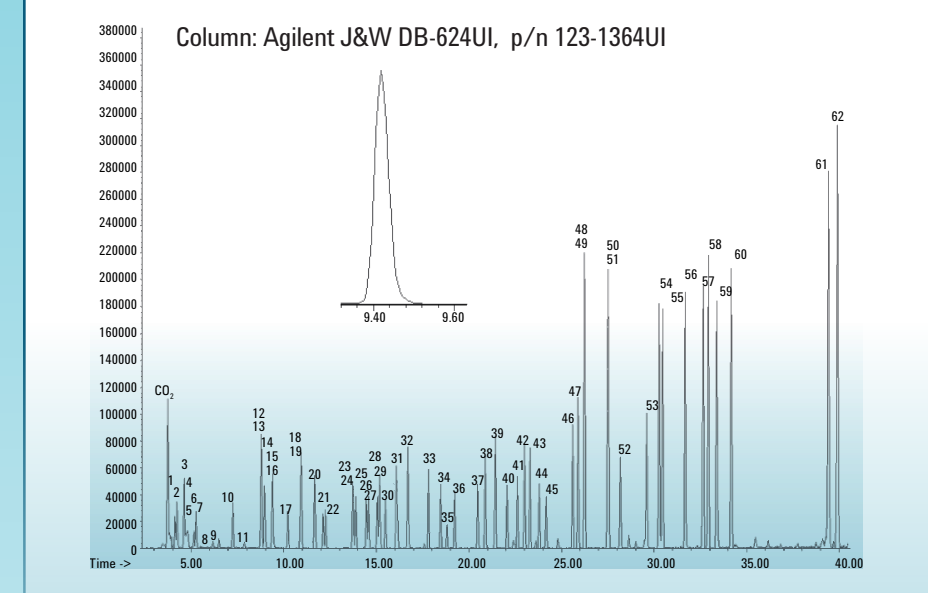
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|------------|--------------------|--------------------|----------------------|----------------------------|
| 1. Ethane | 5. 2-Methylpropane | 9. But-1-ene | 13. Buta-1,3-diene | 17. Isoprene |
| 2. Ethene | 6. n-Butane | 10. cis-But-2-ene | 14. trans-Pent-2-ene | 18. n-Hexane |
| 3. Propane | 7. Acetylene | 11. 2-Methylbutane | 15. Pent-1-ene | 19. Benzene |
| 4. Propene | 8. trans-But-2-ene | 12. Pentane | 16. 2-Methylpentane | 20. 2,2,4-Trimethylpentane |

To see more examples of Ultra volatile analysis, please see Markes Technical note 16. Dual FID dual column set up with UNITY Air Server.



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|------------------|----------------------------|----------------------------|
| 21. n-Heptane | 25. m- & p-Xylene | 29. 1,2,4-Trimethylbenzene |
| 22. Toluene | 26. m- & p-Xylene | 30. 1,2,3-Trimethylbenzene |
| 23. Octane | 27. o-Xylene | |
| 24. Ethylbenzene | 28. 1,3,5-Trimethylbenzene | |

TO-15 air toxics in urban air



Column: Agilent J&W DB-624UI, p/n 123-1364UI
Instrumentation: Markes CIA Advantage with Agilent 7890B GC and 5977A MSD. Splitless analysis of 1 L x 1 ppb air toxics standard using EPA method TO-15 with canister sampling. Note the excellent peak shape in the close-up of extracted mass ion 45 for IPA. See Markes Technical note 81a for method details.

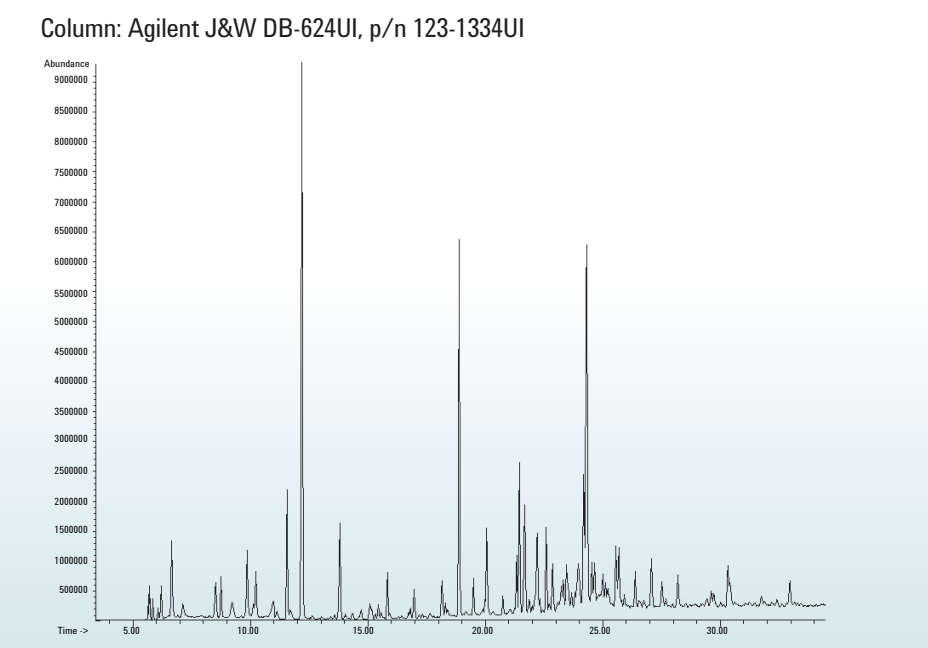
- | | | |
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| 3. 1,2-Dichlorotetrafluoroethane | 24. Ethyl acetate | 46. Chlorobenzene |
| 4. Methyl chloride | 25. Tetrahydrofuran | 47. Ethylbenzene |
| 5. 1,2-Dichloroethane | 26. Chloroform | 48. m-Xylene |
| 6. 1,3-Butadiene | 27. 1,1,1-Trichloroethane | 49. p-Xylene |
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| 11. Ethanol | 32. Trichloroethylene | 54. 1,2,4-Trimethylbenzene |
| 12. 1,2-Dichloroethylene | 33. 1,2-Dichloropropane | 55. 1,3,5-Trimethylbenzene |
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| 19. n-Hexane | 41. 1,1,2-Trichloroethane | |
| 20. 1,1-Dichloroethane | 42. Tetrachloroethylene | |
| 21. Vinyl acetate | 43. Methyl n-butyl ketone | |



INDOOR AIR QUALITY: US EPA METHOD TO-17, EN ISO 16017-1, ASTM D 6196

Most people in the developed world spend an estimated 90% of their time indoors. Regulators and scientists around the world are increasingly concerned about the impact of poor indoor air quality (IAQ) on human health and comfort. Sources of indoor pollutants may range from interior decorations, furnishings and carpets, to construction materials, and even the soil upon which the structure was built. Recent environmental developments (e.g. the EC directive on Energy Performance of Buildings) are putting further pressure on IAQ by reducing building ventilation rates. In this example, pumped tube samplers were used with subsequent TD-GC/MS analysis for profiling of ppt-ppb level VOCs.

Profiling indoor air quality (IAQ)



TD system: Series 2 (ULTRA)UNITY or TD-100
Desorption: 5 min at 280 °C (depends on sorbent)
Trap: To match tube (25 to 300 °C)
Split: During trap desorption only ~15:1
Analysis: GC/MS (scan)

TD-GC/MS analysis of clean indoor air pumped onto a multi-sorbent tube. To find out more about sampling in indoor environments, see Markes Technical TDTS 28.

Agilent 5977/7890B Series GC/MSD and Markes Unity Series 2 TD

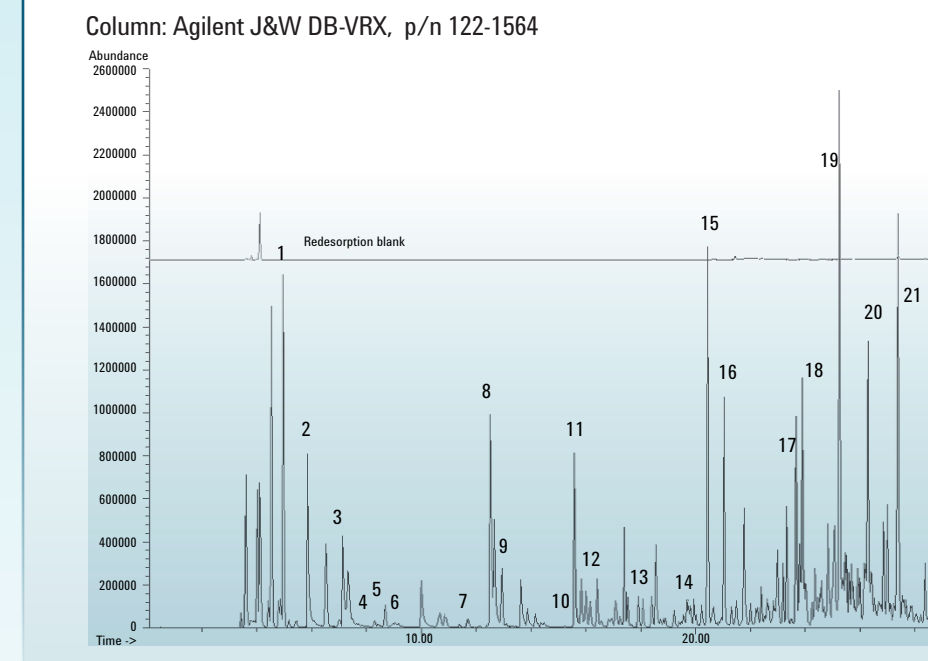


ODOROUS AND TOXIC LANDFILL GAS

The Intergovernmental Panel on Climate Change (IPCC) estimates that 2% of the world's greenhouse gas emissions are caused by landfills.

Consequently, new regulations in Europe and Asia require the monitoring of trace toxic and odorous compounds in landfill gas. These analyses can be performed online, or by active or passive sampling onto sorbent tubes.

Odors and toxics in landfill gas



- | | |
|------------------------------|-------------------------------|
| 1. Vinyl chloride (Toxic) | 12. Benzene |
| 2. Chloroethane | 13. Trichloroethane |
| 3. 1-pentene | 14. Dimethyl disulphide |
| 4. Furan | 15. Toluene |
| 5. Dimethyl sulphide | 16. Butanoic acid ethyl ester |
| 6. Carbon disulphide | 17. Xylene |
| 7. 1,1-dichloroethane | 18. Nonane |
| 8. Butan-2-ol | 19. alpha-pinene |
| 9. 1,1- & 1,2-dichloroethane | 20. Decane |
| 10. 1,1,1-trichloroethane | 21. Limonene |
| 11. Butan-1-ol | |

Trace-level identification of target analytes and major components in 100 mL of landfill gas. The patented inert valve within the Markes Unity Series 2 TD facilitates subsequent offline analysis of the sampled tubes by allowing you to select low flow path temperatures (120 °C in this example). See Markes Technical TDTS 47.

Tubes

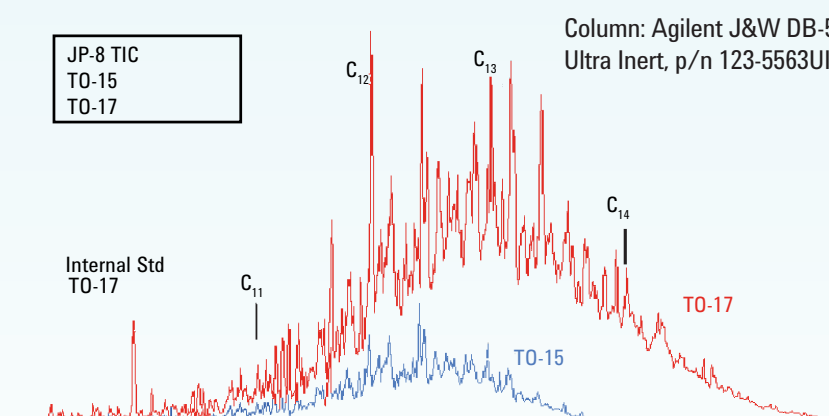
ADVANTAGES	DRAWBACKS
High versatility	Not suitable for the most volatile freons or C_2 hydrocarbons.
Compliance with standard methods, and complete retention of all but the most volatile organic compounds	
Using multiple sorbents, along with backflush desorption, facilitates simultaneous desorption/recovery of analytes of different volatilities	

Tube sampling vs. canister sampling

Given the complexity and variability of organic vapors in air, it is impossible for one sampling approach to suit every monitoring scenario. The two most common sampling strategies include:

- Pumped or diffusive (passive) sampling onto sorbent tubes
- Pumped into a coated stainless steel canister or plastic air tight bag

(Pictured right) Profiles of JP-8 kerosene-contaminated soil gas obtained using canister sampling and TO-15 analysis (blue) and sorbent tube sampling with TO-17 analysis (red). Soil gas measurements assess the risk of vapor intrusion into nearby buildings. See Markes Technical notes TDTS 79 and 80 for method details.



Passivated canisters

ADVANTAGES	DRAWBACKS
Ideal for highly volatile chemicals (such as C_2 hydrocarbons) and non-polar compounds (such as Freons)	Prone to poor recovery of less volatile, or more polar, species
Easy air sampling by releasing a single valve	Require stringent cleaning – including repeated evacuation and purging – between uses
Canisters can be re-used indefinitely	Canisters can be expensive, as well as difficult to transport and store
	TWA sampling is not easy with a canister and requires an elaborate set-up defined by EPA method TO-15

Ensuring an inert GC flow path is critical – and now, easy to achieve

By minimizing flow path activity through proprietary chemistries, Agilent Inert Flow Path solutions ensure accurate quantification and high sensitivity for trace-level analysis.

- Ultra Inert liners – with or without deactivated glass wool – are certified to provide low surface activity and highly reproducible sample vaporization, facilitating delivery of active analytes.
- Inert Inlet weldments are treated to prevent adsorption and degradation.

- Ultra Inert gold-plated inlet seals are manufactured using metal injection molding, gold plating, and application of our Ultra Inert chemistry to produce a leak-free seal that reduces active analyte adsorption.
- Inert MS source ensures sensitivity when analytes reach the mass spectrometer.
- Capillary Flow Technology purged union lets you backflush high boilers in heavy-matrix samples, increasing column lifetime and system productivity.

- UltraMetal Plus Flexible Metal ferrules are the only ferrules that won't introduce active sites into the flow path.
- Agilent J&W Ultra Inert GC columns are rigorously tested to ensure exceptionally low bleed and consistently high inertness for optimal active analyte delivery to the GC or MS detector.
- Gas Clean purifier removes oxygen, moisture, hydrocarbons, and other contaminants.



Don't miss a thing in your GC analysis. Visit agilent.com/chem/inert



To learn more about the best practices for environmental and workplace air monitoring, visit agilent.com/chem/air